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NEWS 5	AUG 24		CA/Caplus enhanced with legal status information for U.S. patents
NEWS 6	SEP 09		50 Millionth Unique Chemical Substance Recorded in CAS REGISTRY
NEWS 7	SEP 11		WPIDS, WPINDEX, and WPIX now include Japanese FTERM thesaurus
NEWS 8	OCT 21		Derwent World Patents Index Coverage of Indian and Taiwanese Content Expanded
NEWS 9	OCT 21		Derwent World Patents Index enhanced with human translated claims for Chinese Applications and Utility Models
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NEWS 12	DEC 01		FRFULL Content and Search Enhancements
NEWS 13	DEC 01		DGENE, USGENE, and PCTGEN: new percent identity feature for sorting BLAST answer sets
NEWS 14	DEC 02		Derwent World Patent Index: Japanese FI-TERM thesaurus added
NEWS 15	DEC 02		PCTGEN enhanced with patent family and legal status display data from INPADOCDB
NEWS 16	DEC 02		USGENE: Enhanced coverage of bibliographic and sequence information
NEWS 17	DEC 21		New Indicator Identifies Multiple Basic Patent Records Containing Equivalent Chemical Indexing in CA/Caplus
NEWS 18	JAN 12		Match STN Content and Features to Your Information Needs, Quickly and Conveniently
NEWS 19	JAN 25		Annual Reload of MEDLINE database
NEWS 20	FEB 16		STN Express Maintenance Release, Version 8.4.2, Is Now Available for Download
NEWS 21	FEB 16		Derwent World Patents Index (DWPI) Revises Indexing of Author Abstracts
NEWS 22	FEB 16		New FASTA Display Formats Added to USGENE and PCTGEN
NEWS 23	FEB 16		INPADOCDB and INPAFAMDB Enriched with New Content and Features
NEWS 24	FEB 16		INSPEC Adding Its Own IPC codes and Author's E-mail Addresses

10/591, 778

NEWS EXPRESS FEBRUARY 15 10 CURRENT WINDOWS VERSION IS V8.4.2,
AND CURRENT DISCOVER FILE IS DATED 15 JANUARY 2010.

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FILE 'HOME' ENTERED AT 08:24:18 ON 27 FEB 2010

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FIEL IS NOT A RECOGNIZED COMMAND  
The previous command name entered was not recognized by the system.  
For a list of commands available to you in the current file, enter  
"HELP COMMANDS" at an arrow prompt (>).
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=> file caplus
COST IN U.S. DOLLARS          SINCE FILE      TOTAL
                                ENTRY        SESSION
FULL ESTIMATED COST          0.22          0.22

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FILE 'CAPLUS' ENTERED AT 08:24:45 ON 27 FEB 2010
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FILE COVERS 1907 - 27 Feb 2010 VOL 152 ISS 10
FILE LAST UPDATED: 26 Feb 2010 (20100226/ED)
REVISED CLASS FIELDS (/NCL) LAST RELOADED: Dec 2009
USPTO MANUAL OF CLASSIFICATIONS THESAURUS ISSUE DATE: Dec 2009

CPlus now includes complete International Patent Classification (IPC) reclassification data for the third quarter of 2009.

CAS Information Use Policies apply and are available at:

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This file contains CAS Registry Numbers for easy and accurate identification.

10/591, 778

substance identification.

```
=> s sterically(l)hindered(l)aminoxide
    19192 STERICALLY
    44142 HINDERED
    93 AMINOXIDE
L1          2 STERICALLY(L)HINDERED(L)AMINOXIDE

=> s sterically(l)hindered(l)amine(l)ether
    19192 STERICALLY
    44142 HINDERED
    316044 AMINE
    575130 ETHER
L2          47 STERICALLY(L)HINDERED(L)AMINE(L)ETHER

=> s 1 or 2
    10554952 1
    10425547 2
SYSTEM LIMITS EXCEEDED - SEARCH ENDED
The search profile you entered was too complex or gave too many
answers. Simplify or subdivide the query and try again. If you have
exceeded the answer limit, enter DELETE HISTORY at an arrow prompt
(=>) to remove all previous answers sets and begin at L1. Use the
SAVE command to store any important profiles or answer sets before
using DELETE HISTORY.

=> s l1 or l2
L3          48 L1 OR L2

=> d 13 1-48 bib abs
```

L3 ANSWER 1 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 2010:147002 CAPLUS

TI Absorbent solution based on a tertiary or hindered amine and on one particular activator and process for removing acid compounds from a gaseous effluent

IN Jacquin, Marc; Grandjean, Julien; Huard, Thierry

PA IFF, Fr.

SO PCT Int. Appl., 32pp.

CODEN: PIXXD2

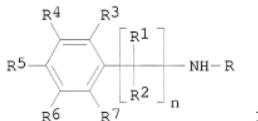
DT Patent

LA French

FAN.CNT 4

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2010012884	A1	20100204	WO 2009-FR902	20090721
	W: AE, AG, AL, AM, AO, AT, AU, AZ, BA, BB, BG, BH, BR, BW, BY, BZ, CA, CH, CL, CN, CO, CR, CU, CZ, DE, DK, DM, DO, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, GT, HN, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KN, KP, KR, KZ, LA, LC, LK, LR, LS, LT, LU, LY, MA, MD, ME, MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PE, PG, PH, PL, PT, RO, RS, RU, SC, SD, SE, SG, SK, SL, SM, ST, SV, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, ZA, ZM, ZW RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HR, HU, IE, IS, IT, LT, LU, LV, MC, MK, MT, NL, NO, PL, PT, RO, SE, SI, SK, SM, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GO, GW, ML, MR, NE, SN, TD, TG, BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM				
	FR 2934172	A1	20100129	FR 2008-4304	20080728
PRAI	FR 2008-4304	A	20080728		

GI



AB The invention relates to the removal of acid compds. in a gaseous effluent in an absorption process that uses an aqueous solution comprising a tertiary amine or a sterically hindered amine as a mixture with a primary or secondary amine corresponding to the general formula (I), where $n = 1$ or 2 , and each of the groups R1, R2, R3, R4, R5, R6, R7 and R is chosen independently from one of the group consisting of: a hydrogen atom, a linear, branched or cyclic C1-12 alkyl group, an aryl group, a hydroxylalkyl group or a linear, branched or cyclic C1-12 ether oxide group. The invention advantageously applies for the treatment of natural gas and of gas of industrial origin.

RE.CNT 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD
 ALL CITATIONS AVAILABLE IN THE RE FORMAT

L3 ANSWER 2 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN

AN 2008:1529938 CAPLUS

DN 150:80828

TI Nitroxide containing electrode materials for secondary batteries

IN Nesvadba, Peter; Bugnon Folger, Lucienne; Hintermann, Tobias

PA Ciba Holding Inc., Switz.

SO PCT Int. Appl., 58pp.

CODEN: PIXXD2

DT Patent

LA English

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI WO 2008155247	A1	20081224	WO 2008-EP57138	20080609
W: AE, AG, AL, AM, AO, AT, AU, AZ, BA, BB, BG, BH, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DO, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, GT, HN, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KN, KP, KR, KZ, LA, LC, LK, LR, LS, LT, LU, LY, MA, MD, ME, MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RS, RU, SC, SD, SE, SG, SK, SL, SM, SV, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, ZA, ZM, ZW RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HR, HU, IE, IS, IT, LT, LU, LV, MC, MT, NL, NO, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG, BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM				

PRAI EP 2007-110574 A 20070619

OS MARPAT 150:80828

AB This invention relates to a stable secondary battery utilizing as active principle the oxidation and reduction cycle of a sterically hindered nitroxide radical, a sterically hindered oxoammonium cation, a sterically hindered hydroxylamine or a sterically hindered aminoxide anion containing a piperazin-2,6-dione, a piperazin-2-one or morpholin-2-one structural unit. Further aspects of the invention are a method for providing such a secondary battery, the use of the resp. compds. as active elements in secondary batteries and selected novel compds.

RE.CNT 10 THERE ARE 10 CITED REFERENCES AVAILABLE FOR THIS RECORD
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L3 ANSWER 3 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 2008:941646 CAPLUS
 DN 149:227491
 TI Desulfurization of natural gas by absorption
 IN Magne-Drisch, Julia; Lucquin, Anne-Claire; Streicher, Christina; Elgue, Jean; Cousin, Jean-Paul; Perdu, Gauthier; Roquet, Damien; Dinh, Viep Hoang
 PA Institut Francais du Petrole, Fr.; Total SA
 SO U.S. Pat. Appl. Publ., 17 pp.
 CODEN: USXXCO

DT Patent
 LA English

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI US 20080187485	A1	20080807	US 2007-671650	20070206
PRAI US 2007-671650		20070206		

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

AB Hydrogen sulfide contained in a hydrocarbon gas, especially natural gas, containing

traces of aromatic hydrocarbons is removed by (a) contacting the gas with an absorbent solution to obtain a gas depleted in H₂S and a H₂S-rich absorbent solution, (b) heating and expanding the H₂S-rich absorbent solution to a predetd. temperature and pressure to release a gaseous fraction containing aromatic

hydrocarbons and to obtain an absorbent solution depleted in aromatic hydrocarbons, with the temperature and pressure being selected such that the gaseous fraction contains at least 50% of the aromatic hydrocarbons and at most 35% hydrogen sulfide contained in the H₂S-rich absorbent solution, (c) thermally regenerating the absorbent solution depleted in aromatic hydrocarbon compds. to release a H₂S-rich gaseous effluent and to obtain a regenerated absorbent solution At least a part of the H₂S-rich gaseous effluent of stage (c) is treated by a Claus process. The absorbent is an aqueous amine solution The amine can be methylidethanolamine, diisopropanolamine, monoethanolamine, diethanolamine, or a sterically hindered amine at a temperature of 80-140° and a pressure of 1.5-6 bara. The absorbent can be sulfolane, methanol, N-formylmorpholine, acetylmorpholine, propylene carbonate, polyethylene glycol di-Me ether, or N-Me pyrrolidone.

L3 ANSWER 4 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 2008:43070 CAPLUS
 DN 148:145644
 TI Process for the preparation of sterically hindered nitroxyl ethers
 IN Basbas, Abdel-Ilah; Alvisi, Davide; Cordova, Robert; DiFazio, Michael
 Peter; Fischer, Walter; Kotrola, Joseph A.; Nocentini, Tiziano; Robbins,
 James; Schoening, Kai-Uwe
 PA Ciba Specialty Chemicals Holding Inc., Switz.
 SO PCT Int. Appl., 38pp.
 CODEN: PIXXD2
 DT Patent
 LA English
 FAN.CNT 2

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2008003605	A1	20080110	WO 2007-EP56301	20070625
	W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BH, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DO, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, GT, HN, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KN, KP, KR, KZ, LA, LC, LK, LR, LS, LT, LU, LY, MA, MD, ME, MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RS, RU, SC, SD, SE, SG, SK, SL, SM, SV, SY, TJ, TM, TN, TR, TT, TZ, UA, US, UZ, VC, VN, ZA, ZM, ZW RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, LV, MC, MT, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG, BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM				
	CA 2654837	A1	20080110	CA 2007-2654837	20070625
	EP 2035382	A1	20090318	EP 2007-786825	20070625
	EP 2035382	B1	20091118		
	R: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LI, LT, LU, LV, MC, MT, NL, PL, PT, RO, SE, SI, SK, TR, AL, BA, HR, MK, RS				
	JP 2009541429	T	20091126	JP 2009-517150	20070625
	AT 449072	T	20091215	AT 2007-786825	20070625
	KR 2009031523	A	20090326	KR 2008-731543	20081226
	CN 101484423	A	20090715	CN 2007-80025374	20090104
	IN 2009CN00058	A	20090605	IN 2009-CNS8	20090105
	MX 2009000047	A	20090123	MX 2009-47	20090107
PRAI	EP 2006-116619	A	20060705		
	EP 2007-106899	A	20070425		
	WO 2007-EP56301	W	20070625		
OS	CASREACT	148:145644;	MARPAT	148:145644	
GI					

* STRUCTURE DIAGRAM TOO LARGE FOR DISPLAY - AVAILABLE VIA OFFLINE PRINT *

AB The present invention relates to a novel process for the preparation of specific sterically hindered nitroxyl ethers from their corresponding sterically hindered nitroxyl radicals by reacting it with an aldehyde and a hydroperoxide. This nitroxyl ether formation may be carried out from different starting nitroxyl radicals, which are subsequently further reacted to the desired compds. The compds. prepared by this process are effective as stabilizers for polymers against harmful effects of light, oxygen and/or heat and as flame-retardants for polymers. Thus, a process for the preparation

of a sterically hindered nitroxyl ether of formula I, wherein n is a number from 1 to 10 and R1 is C1-C5 alkyl, comprises the steps of (a) reacting the nitroxyl compound with n-hexylaldehyde, n-pentylaldehyde, n-butyraldehyde, n-propylaldehyde or acetaldehyde and a hydroperoxide in the presence of a metal catalyst; or (b) reacting 1-oxyl-2,2,6,6-tetramethylpiperidin-4-one with n-hexylaldehyde, n-pentylaldehyde, n-butyraldehyde, n-propylaldehyde or acetaldehyde or a mixture of said aldehydes with their resp. alcs. and a hydroperoxide in the presence of a metal 0 catalyst to yield 1-alkoxy-2,2,6,6-tetramethylpiperidin-4-one, which is further reacted to form a sterically hindered nitroxyl ether of formula I. Thus, a process for the preparation of a sterically hindered nitroxyl ether of formula II, wherein N4-amine is H2N-(CH2)3-NH-(CH2)2-NH-(CH2)3-NH2, is by reacting a compound of formula 1-oxyl-2,2,6,6-tetramethylpiperidin-4-ol or 1-oxyl-2,2,6,6-tetramethylpiperidin-4-one with a compound of formula 1,2,3,6-Tetrahydrobenzaldehyde or cyclohexanecarboxaldehyde and a hydroperoxide in the presence of a metal catalyst to yield the 1-cyclohex-3-enyloxy/1-cyclohexyloxy of 2,2,6,6-tetramethylpiperidin-4-one/ol, which are further reacted to form a compound of formula II.

L3 ANSWER 5 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 2007:435706 CAPLUS
 DN 146:422837
 TI Production of sterically hindered amine ethers useful as stabilizers for polymer compositions

IN Frey, Markus; Rast, Valerie
 PA Ciba Specialty Chemicals Holding Inc., Switz.
 SO PCT Int. Appl., 89pp.
 CODEN: PIXXD2

DT Patent
 LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2007042422	A2	20070419	WO 2006-EP66962	20061002
	WO 2007042422	A3	20070607		

W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH,
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 GE, GH, GM, HN, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KN, KP,
 KR, KZ, LA, LC, LK, LR, LS, LT, LU, LV, LY, MA, MD, MG, MK, MN,
 MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RS,
 RU, SC, SD, SE, SG, SK, SL, SM, SV, SY, TJ, TM, TN, TR, TT, TZ,
 UA, UG, US, UZ, VC, VN, ZA, ZM, ZW
 RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE,
 IS, IT, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ,
 CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG, BW, GH,
 GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY,
 KG, KZ, MD, RU, TJ, TM, AP, EA, EP, OA

EP 1943236	A2	20080716	EP 2006-806923	20061002
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JP 2009514803	T	20090409	JP 2008-534986	20061002
US 20090318592	A1	20091224	US 2009-83052	20090304

PRAI EP 2005-109432 A 20051011
 WO 2006-EP66962 W 20061002

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

OS MARPAT 146:422837

AB The invention relates to processes for preparation of sterically hindered amine ethers by reacting a sterically hindered amine oxide with a ketone or an aldehyde having at least one reactive H in the presence of a peroxydisulfate. The products obtained may be hydrogenated. The compds. made by these processes are particularly effective in stabilization of polymer compns. against harmful effects of light, oxygen, heat, and as flame-retardants for polymers. Thus, 1-[(2,2,6,6-tetramethyl-1-piperidinyl)oxy]-2-propanone was prepared in 82% (66% isolated) yield by adding TEMPO (7.8) and acetone (30.5) to a stirred solution of silver nitrate (0.17 g) in deionized water (30 mL) at 25°, bringing the mixture to reflux (62°), slowly adding a solution of sodium peroxydisulfate (11.9 g) in water (45 mL) over 3 h, cooling the mixture to 255°, and neutralizing with sodium bicarbonate.

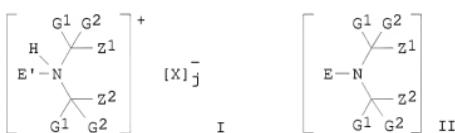
L3 ANSWER 6 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 2006:823408 CAPLUS
 DN 145:250371
 TI Stabilization of natural composites containing cellulosic or wood based fillers with sterically hindered amine
 IN Kaspers, Sarah R.; Guckel, Christian; Rogez, Daniel; Schaller, Christian M.
 PA Ciba Specialty Chemicals Corp., USA
 SO U.S. Pat. Appl. Publ., 37 pp.
 CODEN: USXXCO
 DT Patent
 LA English
 FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI US 20060183821	A1	20060817	US 2006-352873	20060213
WO 2006087269	A1	20060824	WO 2006-EP50673	20060206
W: AB, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KN, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, LY, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW				
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DE 112006000348	T5	20071227	DE 2006-112006000348	20060206
JP 2008531324	T	20080814	JP 2007-555572	20060206
PRAI US 2005-653170P	P	20050215		
WO 2006-EP50673	W	20060206		

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

OS MARPAT 145:250371

GI



AB A method of protecting natural composites against light-induced degradation comprises the steps of: treating cellulosic or wood based fillers with an impregnating composition wherein the impregnating composition comprises (a) water or
 water/organic solvent and (b) a sterically hindered amine which is selected from at least one of the compds. of formulas I and II, wherein G1 and G2 are independently alkyl of 1 to 4 carbon atoms or are together pentamethylene; Z1 and Z2 are each Me, or Z1 and Z2 together form an unsubstituted linking moiety or a linking moiety substituted by one or more groups selected from an ester, ether,

hydroxy, oxo, cyanohydrin, amide, amino, carboxy or an urethane group; E is oxyl, hydroxyl, hydrogen or methyl; E' is hydroxyl, hydrogen or methyl; X is an inorg. or organic acid; and, wherein the total charge of h cations is equal to the total charge of j anions; drying the impregnated fillers, and blending the impregnated fillers with a polymer via compounding methods such as extrusion to form the natural composite.

L3 ANSWER 7 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 2006:440107 CAPLUS
 DN 144:468916
 TI Process for the synthesis of sterically hindered N-alkoxyamines
 IN Frey, Markus; Rast, Valerie; Martinez, Francisco; Alvisi, Davide
 PA Ciba Specialty Chemicals Holding Inc., Switz.
 SO PCT Int. Appl., 75 pp.
 CODEN: PIXXD2
 DT Patent
 LA English
 FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI WO 2006048389	A1	20060511	WO 2005-EP55472	20051024
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, LY, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW				
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EP 1807395	A1	20070718	EP 2005-801650	20051024
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CN 101048378	A	20071003	CN 2005-80037093	20051024
JP 2008519003	T	20080605	JP 2007-539568	20051024
US 20090069470	A1	20090312	US 2007-665885	20070419
PRAI EP 2004-105456	A	20041102		
WO 2005-EP55472	W	20051024		
ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT				
OS MARPAT 144:468916				
AB The present invention relates to novel processes for the preparation of a sterically hindered amine ethers by the transformation of a corresponding oxo-piperidin to a hydroxy or amino substituted sterically hindered amine ether and the preparation of a N-propoxy or N-propenoxy substituted sterically hindered amine and some novel compds. obtainable by these processes. The compds. made by these processes are particularly effective in the stabilization of polymer compns. against harmful effects of light, oxidation and/or heat and as flame-retardants for polymers.				
OSC.G 3 THERE ARE 3 CAPLUS RECORDS THAT CITE THIS RECORD (4 CITINGS)				
RE.CNT 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD				
ALL CITATIONS AVAILABLE IN THE RE FORMAT				

L3 ANSWER 8 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 2005:1042220 CAPLUS
 DN 143:347055

TI A process for the synthesis of sterically hindered amine ethers useful as stabilizing and fireproofing agents

IN Frey, Markus; Rast, Valerie; Braig, Adalbert; Kramer, Andreas

PA Ciba Specialty Chemicals Holding Inc., Switz.

SO PCT Int. Appl., 71 pp.
 CODEN: PIXXD2

DT Patent

LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2005090307	A1	20050929	WO 2005-EP50995	20050307
	W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW				
	RW: BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				
	DE 112005000536	T5	20070201	DE 2005-112005000536	20050307
	JP 2007529467	T	20071025	JP 2007-503327	20050307
	US 20070191516	A1	20070816	US 2006-591778	20060906
PRAI	EP 2004-101047	A	20040315		
	WO 2005-EP50995	W	20050307		

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

OS CASREACT 143:347055; MARPAT 143:347055

AB A process for the preparation of a sterically hindered amine ether which comprises reacting a corresponding sterically hindered aminoxide with a C5-C18alk-1-ene in the presence of an organic hydroperoxide and optionally hydrogenating the resulting product as well as the product mixts. obtained therewith and their use as stabilizers and flame retardants.

OSC.G 2 THERE ARE 2 CAPLUS RECORDS THAT CITE THIS RECORD (5 CITINGS)

RE.CNT 10 THERE ARE 10 CITED REFERENCES AVAILABLE FOR THIS RECORD

ALL CITATIONS AVAILABLE IN THE RE FORMAT

L3 ANSWER 9 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 2005:601518 CAPLUS
 DN 143:99285
 TI Sulphur scavenging amines being monomeric adducts of a sterically hindered amine and an aldehyde or donor thereof

IN Gatlin, Larry W.

PA Clearwater International LLC., USA
 SO Brit. UK Pat. Appl., 39 pp.

CODEN: BAXXDU

DT Patent

LA English

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI GB 2409859	A	20050713	GB 2005-156	20050106
GB 2409859	B	20090401		
US 20050153846	A1	20050714	US 2004-754487	20040109
US 7517447	B2	20090414		
CA 2491973	A1	20050709	CA 2005-2491973	20050107
NO 200500101	A	20050711	NO 2005-101	20050107
AU 2005200048	A1	20050728	AU 2005-200048	20050107
US 20090250659	A1	20091008	US 2009-419418	20090407
PRAI US 2004-754487	A	20040109		

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

OS CASREACT 143:99285

AB Oil-soluble, sulfur scavengers or converts are disclosed where the scavengers include substantially monomeric aldehyde-amine adducts from the reaction of at least one sterically hindered primary or secondary amine and a molar excess of at least one aldehyde or a donor thereof. Preferably the adduct is a compound of formulas CH2RNR1R2, CH2RR4NR3NR5CH2R or mixts. thereof: where R is a H or a carbon-containing group, R1 and R2 are the same or different, at least one being a sterically hindered carbon-containing group having between about 3 and about 24 carbon atoms or R1 and R2 can form a ring system, R3 is a divalent sterically hindered carbon-containing group, R4 and R5 are the same or different and are H or a CH2R group and where one or more of the carbon atoms of R, R1, R2, R3, R4, R5 or mixts. thereof can be replaced by oxygen atoms in the form of ether moieties, nitrogen groups in the form of tertiary amine or amide moieties or mixts. thereof, and where one or more hydrogen atoms of R, R1, R2, R3, R4, R5 or mixts. thereof can be replaced by fluorine atoms, chlorine atoms or mixts. thereof. Methods are also disclosed for reducing, reducing below a given level or eliminating noxious sulfur species from fluids using these scavengers and for making them. A plot shows pressure and H2S concentration verses time for a scavenger embodying this invention compared with comparative triazine scavengers tested at high pressure in a sour gas-drilling mud system pressure and headspace H2S composition profiles at about a 10:1 scavenger to H2S ratio.

OSC.G 1 THERE ARE 1 CAPLUS RECORDS THAT CITE THIS RECORD (1 CITINGS)

RE.CNT 6 THERE ARE 6 CITED REFERENCES AVAILABLE FOR THIS RECORD

ALL CITATIONS AVAILABLE IN THE RE FORMAT

L3 ANSWER 10 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 2004:355004 CAPLUS
 DN 140:340168
 TI Flame retardant compositions containing sterically hindered amine stabilizers

IN Kaprinidis, Nikolas; Lelli, Nicola
 PA Ciba Specialty Chemicals Holding Inc., Switz.
 SO PCT Int. Appl., 101 pp.
 CODEN: PIXXD2

DT Patent
 LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2004035671	A1	20040429	WO 2003-EP11185	20031009
	W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, UZ, VC, VN, YU, ZA, ZM, ZW				
	RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				
US	20040057620	A1	20040520	US 2003-675157	20030930
US	7109260	B2	20060919		
CA	2501384	A1	20040429	CA 2003-2501384	20031009
AU	2003282026	A1	20040504	AU 2003-282026	20031009
EP	1554336	A1	20050720	EP 2003-773639	20031009
	R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, SK				
CN	1705705	A	20051207	CN 2003-80101433	20031009
CN	100439434	C	20081203		
JP	2006503138	T	20060126	JP 2004-544120	20031009
TW	291478	B	20071221	TW 2003-92128535	20031015
PRAI	US 2002-419260P	P	20021017		
	WO 2003-EP11185	W	20031009		

OS MARPAT 140:340168

AB Organic polymeric substrates, for example polyolefins such as polypropylene, can be made flame retardant by the incorporation of a synergistic mixture of (i) at least one sterically hindered amine stabilizer, (ii) at least one conventional flame retardant selected from the group consisting of the organohalogen, phosphorous containing, isocyanurate and melamine based flame retardants and (iii) at least one acid scavenger. The compns. of the invention combine good flame retardant properties with light stability and good mech. properties. Polyolefin molded articles are stabilized against light, heat and oxygen and made flame retardant with the incorporation of at least one sterically hindered amine and at least one conventional flame retardant, while allowing normally high levels of flame-retardant fillers to be greatly reduced or eliminated.

OSC.G 6 THERE ARE 6 CAPLUS RECORDS THAT CITE THIS RECORD (6 CITINGS)

RE.CNT 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD

ALL CITATIONS AVAILABLE IN THE RE FORMAT

L3 ANSWER 11 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
AN 2004:108785 CAPLUS
DN 141:332890
TI Flame retardant and UV stabilized compositions
AU Anon.
CS Switz.
SO IP.com Journal (2003), 3(11), 5 (No. IPCOM000019831D), 1 Oct 2003
CODEN: IJPOBX; ISSN: 1533-0001
PB IP.com, Inc.
DT Journal; Patent
LA English
FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	IP 19831D		20031001	IP 2003-19831D	20031001
PRAI	IP 2003-19831D		20031001		
AB A TPO-TPE-TPV (automotive) skin, which can be extruded, calendered, slush-molded or thermoformed, composed of a polyolefin resin and a sterically hindered amine ether flame retardant exhibits excellent flame retardant properties and UV resistance. Another example for combining excellent properties in respect of flame retardancy and UV stability is an expanded polyolefin such as polyethylene (EPE) or expanded polypropylene (EPP) or expanded polyolefin foam, which can be extruded, comprising a sterically hindered amine ether.					

L3 ANSWER 12 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
AN 2003:631108 CAPLUS
TI Selective oxidation of 2,6-dimethylphenol to polyphenylene ether using copper coordination compounds
AU Reedijk, Jan; Gamez, Patrick
CS Leiden Institute of Chemistry, Leiden University, Leiden, 2300 RA, Neth.
SO Abstracts of Papers, 226th ACS National Meeting, New York, NY, United States, September 7-11, 2003 (2003), COLL-305 Publisher: American Chemical Society, Washington, D. C.
CODEN: 69EKY9
DT Conference; Meeting Abstract
LA English
AB The engineering plastic polyphenylene ether (PPE) is commonly produced from 2,6-dimethylphenol and dioxygen, under alkaline conditions with Cu amine compds. as catalysts (see Figure); also many other oxidns. catalyzed by copper coordination compds. and in biol. Cu systems are known. The major technol. problems in the PPE process are the highly colored side product diphenoquinone, water poisoning and the control of the mol. weight. Using sterically hindered amine ligands, proper solvent choice and bilayered solvents, these problems have been tackled and reduced. A short overview of the most recent findings and the mechanistic implications will be discussed.

L3 ANSWER 13 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 2003:435329 CAPLUS
 DN 139:7697
 TI Hindered amine ether flame retardant polyolefin resin pre-expanded particles and in-mold foamed articles prepared therefrom
 IN Stuart, John Bready; Skarke, Steven Charles; Ogita, Tetsuya; Yamaguchi, Takema; Iwamoto, Tomonori
 PA Kaneka Corporation, Japan; Kaneka Texas Corporation
 SO U.S. Pat. Appl. Publ., 12 pp.
 CODEN: USXXCO
 DT Patent
 LA English
 FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 20030105194	A1	20030605	US 2001-998170	20011203
	US 6822023	B2	20041123		
	WO 2003048239	A1	20030612	WO 2002-JP12565	20021129
	W: CN, JP, SG RW: BE, DE, FR, GB				
	EP 1454947	A1	20040908	EP 2002-783720	20021129
	EP 1454947	B1	20100217		
	R: BE, DE, FR, GB				
	CN 1599771	A	20050323	CN 2002-824180	20021129
	CN 1309768	C	20070411		
PRAI	US 2001-998170	A	20011203		
	WO 2002-JP12565	W	20021129		

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

AB Flame retardant polyolefin pre-expanded particles made of a resin composition comprising a polyolefin resin and a sterically hindered amine ether flame retardant, which can be molded with good moldability to give in-mold foamed articles which have an excellent flame resistance and do not generate harmful gas at the time of burning.

OSC.G 1 THERE ARE 1 CAPLUS RECORDS THAT CITE THIS RECORD (1 CITINGS)

RE.CNT 2 THERE ARE 2 CITED REFERENCES AVAILABLE FOR THIS RECORD

ALL CITATIONS AVAILABLE IN THE RE FORMAT

L3 ANSWER 14 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
AN 2002:300134 CAPLUS
DN 137:170290
TI A molecular study towards the interaction of phenolic anti-oxidants,
aromatic amines and HALS stabilizers in a thermo-oxidative ageing process
AU Barret, Julien; Gijsman, Pieter; Swagten, Josefien; Lange, Ronald F. M.
CS DSM Research, Geleen, 6160 MD, Neth.
SO Polymer Degradation and Stability (2002), 76(3), 441-448
CODEN: PDSTDW; ISSN: 0141-3910
PB Elsevier Science Ltd.
DT Journal
LA English
AB The mol. interactions of phenolic antioxidants, aromatic amines, and HALS
stabilizers are studied by heating the stabilizers under oxidative
conditions in polar and non-polar solvents. The polar solvent
bis(2-methoxyethyl) ether is used to mimic polar engineering
plastics like e.g. TPE-U's, whereas the non-polar solvent squalane or
2,6,10,15,19,23-hexamethyltetrasacosane is used to mimic polypropylene. The
oxidation rate is followed by the anal. of samples taken in time using
various anal. techniques as e.g. IR, HPLC-PDA, GC-FID, GC-MS, and LC-MS.
A general occurring interaction between sterically
hindered phenols and aromatic amines, i.e. regeneration of the aromatic
amine by the sterically hindered phenol, is
demonstrated by varying the mol. structure of the phenol as well as the
polarity of the system. Studies using mixts. of a HALS with structurally
different sterically hindered phenols visualized a
general antagonistic effect between the phenols and the HALS, in which the
HALS consumes the phenol. In all stabilizer combinations using the
sterically hindered phenols studied here, a
stabilization of the resulting quinone form is observed which can, dependent
on the mol. structure of this quinone, lead to solubility issues in polymers.
OSC.G 11 THERE ARE 11 CAPLUS RECORDS THAT CITE THIS RECORD (11 CITINGS)
RE.CNT 12 THERE ARE 12 CITED REFERENCES AVAILABLE FOR THIS RECORD
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L3 ANSWER 15 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
AN 2002:1646 CAPLUS
DN 136:310551
TI The interaction of a phenolic anti-oxidant and an aromatic amine in a thermo-oxidative ageing process
AU Barret, Julien; Gijsman, Pieter; Swagten, Josefien; Lange, Ronald F. M.
CS DSM Research, Geleen, 6160 MD, Neth.
SO Polymer Degradation and Stability (2002), 75(2), 367-374
CODEN: PDSTDW; ISSN: 0141-3910
PB Elsevier Science Ltd.
DT Journal
LA English
AB An interaction of the sterically hindered phenolic antioxidant 1,3,5-trimethyl-2,4,6-Tris(3,5-di-t-butyl-4-hydroxybenzyl)-benzene, 1a, with the aromatic amine 4,4'-bis(α,α -dimethyl-benzyl)diphenylamine, 2a, in co-poly(ether esters) has been observed. This interaction is successfully analyzed by heating 1a and/or 2a in the polyether-mimicking solvent bis(2-methoxyethyl)ether using various anal. techniques. As well as a detailed insight of the action of the individual antioxidants, it has been shown that the quinone structure of the oxidized sterically hindered phenolic antioxidants are only stable in the absence of radical species, which results for the 1a-2a system in the stabilization of the oxidized 1a.
OSC.G 4 THERE ARE 4 CAPLUS RECORDS THAT CITE THIS RECORD (4 CITINGS)
RE.CNT 16 THERE ARE 16 CITED REFERENCES AVAILABLE FOR THIS RECORD
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L3 ANSWER 16 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 2001:886344 CAPLUS
 DN 136:38282
 TI Aging-resistant epoxy-resin casting compositions, molded articles and their use
 IN Bogner, Georg; Brunner, Herbert; Hoesch, Klaus
 PA Osram Opto Semiconductors GmbH & Co. Ohg, Germany
 SO PCT Int. Appl., 26 pp.
 CODEN: PIXXD2
 DT Patent
 LA German
 FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2001092396	A1	20011206	WO 2001-DE1848	20010516
	W: CA, CN, JP, KR, US RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR				
	DE 10027206	A1	20011213	DE 2000-10027206	20000531
	EP 1287063	A1	20030305	EP 2001-944932	20010516
	EP 1287063	B1	20050316		
	R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR				
	CN 1432040	A	20030723	CN 2001-810252	20010516
	CN 1216933	C	20050831		
	JP 2003553170	T	20031125	JP 2002-500599	20010516
	CN 1706886	A	20051214	CN 2005-10085912	20010516
	TW 260331	B	20060821	TW 2001-90112884	20010529
	US 20030144414	A1	20030731	US 2002-296848	20021127
	US 7183661	B2	20070227		
PRAI	DE 2000-10027206	A	20000531		
	CN 2001-810252	A3	20010516		
	WO 2001-DE1848	W	20010516		

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

OS MARPAT 136:38282

AB A 2-component casting and potting resin composition for assembling and encapsulating automotive, electronic and optoelectronic parts, modules and components for outdoor use, especially LEDs, and suitable for surface mount technol. (SMT), comprises resin component A containing sterically hindered amine as photochem. aging stabilizer, and curing component B which is an acid anhydride hardener. For example, a title composition cured with partially esterified methylhexahydrophthalic anhydride comprised Rutapox VE 3748 (a bisphenol A diglycidyl ether), Byk A-506, Silane A-187 (adhesion enhancer), optical brightener and 0.2% Tinuvin 123 (photochem. stabilizer).

OSC.G 3 THERE ARE 3 CAPLUS RECORDS THAT CITE THIS RECORD (3 CITINGS)

RE.CNT 4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD

ALL CITATIONS AVAILABLE IN THE RE FORMAT

L3 ANSWER 17 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 2000:470611 CAPLUS
 DN 133:135742

TI Process for manufacture of stable polyolefins
 IN Zhang, Baoqing; Jing, Zhenhua; Hong, Xiaoyu
 PA China Petrochemical Corp., Peop. Rep. China
 SO Faming Zhuanli Shenqing Gongkai Shuomingshu, 9 pp.
 CODEN: CNXXEV

DT Patent
 LA Chinese

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI CN 1231292	A	19991013	CN 1998-101246	19980403
CN 1115349	C	20030723		

PRAI CN 1998-101246

OS MARPAT 133:135742

AB The stable polyolefin is produced by catalytic polymerization of C5-12 α -olefin in the presence of solid catalyst, alkylaluminum co-catalyst, phenol-type antioxidant, and/or sterically hindered amine. The solid catalyst is composed of TiCl₄, 8.0-15.0% ester and ether internal electron donor compound, and MgCl₂ carrier; the contents of Mg and Ti in the solid catalyst are 15-25% and 1.0- 5.0%, resp., and the ratio of ether to ester is 0.6-2.8. The ester is R₃OOC-C(R₁)=C(R₂)-COOR₄ (R₁-R₄ = C₁-10 alkyl, aryl, or arylalkyl), preferably diisobutyl phthalate, and the ether is R₃₀-CH₂-C(R₁)(R₂)-CH₂-OR₄, preferably 2,2-di-tert-butyl-1,3-dimethoxypropane. R₁R₂Si(OR)₂ external electron donor compound may be added in the polymerization process, and the mole ratio

of Al to Si is 30-40. The phenol-type antioxidant is 2,3,5,6-tetra(R)-4-R'-phenol (R = H, C₁-30 alkyl, C₃-30 cycloalkyl, C₆-30 aryl, C₇-30 arylalkyl, or C₁-30 alkoxy; and R' = CHR₂, or OR), preferably pentaerythritol tetrakis[3-(3,5-di-tert-butyl-4-hydroxyphenyl)propanoate], and the ratio of antioxidant to the polymer is 0.01-0.9%. The ratio of photostabilizer to the polymer is 0.01-0.9%.

L3 ANSWER 18 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 2000:387633 CAPLUS
 DN 133:164354
 TI Synthesis of Novel Hindered Amine Light Stabilizers (HALS) and Their Copolymerization with Ethylene or Propylene over Both Soluble and Supported Metallocene Catalyst Systems
 AU Wilen, Carl-Eric; Auer, Markku; Strandén, Juha; Naesman, Jan H.; Rotzinger, Bruno; Steinmann, Alfred; King, Roswell E., III; Zweifel, Hans; Drewes, Rolf
 CS Laboratory of Polymer Technology, Abo Akademi University, Turku, FIN-20500, Finland
 SO Macromolecules (2000), 33(14), 5011-5026
 CODEN: MAMOBX; ISSN: 0024-9297
 PB American Chemical Society
 DT Journal
 LA English
 AB Novel polymerizable hindered amine light stabilizers (HALS) such 1-(but-3-enyl)-2,2,6,6-tetramethylpiperidine (1), 1-(undec-10-enyl)-2,2,6,6-tetramethylpiperidine (2), 4-(but-3-enyl)-1,2,2,6,6-pentamethyl-3,4-dehydropiperidine (3), 2-(but-3-enyl)-2,6,6-trimethylpiperidine (4), 4-(but-3-enyl)-1,2,2,6,6-pentamethyl-4-piperidyl ether (5), 4-(undec-10-enylamide)-1,2,2,6,6-pentamethylpiperidine (6), 4-(N-n-butyl)-undec-10-enylamide)-1,2,2,6,6-pentamethylpiperidine (7), and bis(N-n-butyl-N-2,2,6,6-tetramethylpiperidine)-N-n-butyl-N-allyltriazine (8) were synthesized. All the aforementioned HALS monomers except for 5 and 8 were successfully copolymerized in fair to high yields with ethylene or propylene over 8 different group 4 metallocene catalysts using methylalumoxane (MAO) as cocatalyst. Copolymers were also performed over a supported metallocene/SiO₂/MAO/triisobutylaluminum(TIBA) catalyst system. The silica-supported metallocene catalyst system readily promoted copolymerization of the sterically hindered monomer 2 with ethylene, although copolymers using either 6 or 7 as comonomer failed. Moreover, a catalyst derived from the reaction of rac-[dimethylsilylenebis(1-indenyl)]zirconium dichloride (CA1) with Et₃Al and trityl tetra(perfluorophenyl)borate (TRI-FABA) afforded HALS copolymers in high yields. Surprisingly, TRI-FABA, a strong Lewis acid, could impede the Lewis base activity of HALS monomers such as 2, 6 and 7 provided a sufficient relative amount of TRI-FABA was employed. Thus, once an equilibrium concentration between TRI-FABA and HALS monomer was established, the presence of HALS monomer no longer affected the rate of polymerization. Normally, metallocene catalysts are severely poisoned when traces of polar monomers (Lewis bases) are present, due to the Lewis acidic nature of the catalyst. Furthermore, a series of standard ethylene homopolymers over rac-[dimethylsilylenebis(4,5,6,7-tetrahydro-1-indenyl)]zirconium dichloride (CA2)/MAO catalyst system was performed in the presence of different sterically hindered amine model compds. such as 1-(1-methlene-2,6-di-tert-butylphenol)-2,2,6,6-tetramethylpiperidine (A), N,N-disopropylaniline (B), 1-octyl-2,2,6,6-tetramethylpiperidine (C), 1-benzyl-2,2,6,6-tetramethylpiperidine (D), 2,2,6,6-tetramethylpiperidine (E), 1,2,2,6,6-pentamethylpiperidine (F), diisopropylethylamine (G), 1,2,2,6,6-pentamethyl-4-oxopiperidine (H), 2,2,6,6-tetramethylpiperidine-1-oxyl (I), 1-propargyl-2,2,6,6-tetramethylpiperidine (J), 4-N,N-bis(n-butylamino)-2,2,6,6-tetramethylpiperidine (K), tris(N-butyl-N-2,2,6,6-tetramethylpiperidineamino)triazine (L), and tris(dibutylamino)triazine (M). Some of the amine model compds.

are highly reactive and deactivating whereas others are less so. Much preferred are those HALS structures which have a sterically demanding substituent attached on N and no addnl. heteroatoms in 4-position of the piperidine ring in terms of metallocene/MAO catalyst activity. ¹³C NMR analyses revealed that the produced materials are random copolymers containing isolated HALS branches and that the propylene copolymers have highly stereoregular microstructures. According to size exclusion chromatog., the copolymers have mol. weight distributions close to 2, which are characteristic for polymers produced over single-site catalysts. The copolymers contained 0.2-14.1 weight-% of HALS units and exhibited high UV and thermooxidative stabilities even after exhaustive extraction with a mixture of refluxing (50:50) cyclohexane/2-propanol. For example, the poly(ethylene-co-4) copolymer with a HALS content of 0.2 weight-% exhibited considerable improved thermooxidative stability in comparison to unstabilized polyethylene, i.e., for the copolymer the carbonyl peak had not appeared after one year of oven aging at 115°, whereas unstabilized polyethylene shows a strong increase in the carbonyl index within 2 days.

OSC.G 14 THERE ARE 14 CAPLUS RECORDS THAT CITE THIS RECORD (14 CITINGS)
RE.CNT 63 THERE ARE 63 CITED REFERENCES AVAILABLE FOR THIS RECORD
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L3 ANSWER 19 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
AN 2000:371854 CAPLUS
DN 132:335883
TI Compositions containing acrylic and indene resins for coatings for wood,
masonry, and metals

IN Jurcau, Dorin; Gherdan, Mircea; Tamas, Ladislau; Drutau, Mihaela
PA S.C. "Azur" S.A., Timisoara, Rom.

SO Rom., 4 pp.
CODEN: RUXXA3

DT Patent
LA Romanian

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI RO 110523	B1	19960130	RO 1991-146941	19910218

PRAI RO 1991-146941 19910218

AB Glossy coatings with good resistance to water and salt solns. for the
title substrates are manufactured from compns. containing 50% acrylic
resin-PhMe
solution 15-35, pigment and fillers 15-32, pigment dispersants 0.2-2, 50%
indene resin-PhMe solution 30-60, BuOAc 1-4, ethylene glycol monoethyl
ether acetate 0.3-3, solvent naphtha 1-50, and sterically
hindered amine and oxalamide UV absorber 0.1-1 parts.

The acrylic resin is manufactured by polymerization of Me methacrylate 3-25,
styrene

20-60, Bu acrylate 10-50, Bu methacrylate 3-15, 2-ethylhexyl acrylate
10-40, α -methylstyrene 3-20, and acrylic acid 1-10 parts.

L3 ANSWER 20 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN

AN 1999:205371 CAPLUS

DN 130:237281

TI Preparation of fluoromethyl ethers by the fluorination of chloromethyl ethers with sterically hindered tertiary amine hydrofluoride salts

IN Kudzma, Linas V.; Lessor, Ralph A.; Rozov, Leonid A.; Ramig, Keith

PA Baxter International Inc., USA

SO U.S., 4 pp.

CODEN: USXXAM

DT Patent

LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 5886239	A	19990323	US 1997-975689	19971121
	TW 472035	B	20020111	TW 1998-87118992	19981117
CA	2277928	A1	19990603	CA 1998-2277928	19981118
CA	2277928	C	20080610		
WO	9926910	A1	19990603	WO 1998-US24746	19981118
	W: AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ, DE, DK, EE, ES, FI, GB, GE, GH, GM, HR, HU, ID, IL, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, UA, UG, UZ, VN, YU, ZW				
	RW: GH, GM, KE, LS, MW, SD, SZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG				
AU	9914205	A	19990615	AU 1999-14205	19981118
AU	752881	B2	20021003		
EP	958269	A1	19991124	EP 1998-958096	19981118
	R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, NL, SE, PT, IE, FI				
BR	9806786	A	20000509	BR 1998-6786	19981118
HU	2000003190	A2	20010528	HU 2000-3190	19981118
HU	2000003190	A3	20011128		
JP	2001508468	T	20010626	JP 1999-529114	19981118
JP	4355369	B2	20091028		
RU	2169724	C2	20010627	RU 1999-118578	19981118
EP	1277724	A1	20030122	EP 2002-79024	19981118
EP	1277724	B1	20050629		
	R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, NL, SE, PT, IE, FI				
CN	1140488	C	20040303	CN 1998-801928	19981118
AT	298738	T	20050715	AT 2002-79024	19981118
ES	2241953	T3	20051101	ES 2002-79024	19981118
PL	191789	B1	20060731	PL 1998-334702	19981118
ZA	9810627	A	19990525	ZA 1998-10627	19981120
MX	9906614	A	20000228	MX 1999-6614	19990715
PRAI	US 1997-975689	A	19971121		
	EP 1998-958096	A3	19981118		
	WO 1998-US24746	W	19981118		

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

OS CASREACT 130:237281; MARPAT 130:237281

AB Fluoromethyl ethers R1(R2)C(A)OCH₂F [A = lower fluoroalkyl, fluorine, chlorine; R1, R2 = hydrogen, (un)branched lower alkyl, lower fluoroalkyl, fluoro, chloro; such that ≥1 of A, R1, R2 = lower fluoroalkyl, (un)branched lower alkyl (e.g., 2,2,2-trifluoro-1-(trifluoromethyl)ethyl ether)], useful as inhalation anesthetics, are prepared by the fluorination of the corresponding chloromethyl ether (e.g., 2-(chloromethoxy)-1,1,1,3,3-hexafluoropropanol with a sterically hindered tertiary amine hydrofluoride salt (e.g., diisopropylethylamine hydrofluoride).

10/591,778

OSC.G 8 THERE ARE 8 CAPLUS RECORDS THAT CITE THIS RECORD (8 CITINGS)
RE.CNT 17 THERE ARE 17 CITED REFERENCES AVAILABLE FOR THIS RECORD
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L3 ANSWER 21 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 1998:701048 CAPLUS
 DN 129:317691
 OREF 129:64797a

TI Polyurethane coating material, method for its preparation and application as coating varnish or clear varnish, especially for coating of synthetic materials

IN Hintze-Bruenning, Horst
 PA Basf Coatings A.-G., Germany
 SO Ger., 16 pp.
 CODEN: GWXXAW

DT Patent
 LA German

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	DE 19723504	C1	19981015	DE 1997-19723504	19970605
CA	2289510	A1	19981210	CA 1998-2289510	19980602
WO	9855526	A1	19981210	WO 1998-EP3285	19980602
	W: BR, CA, JP, KR, US RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE				
EP	986595	A1	20000322	EP 1998-929411	19980602
EP	986595	B1	20020417		
	R: AT, BE, CH, DE, DK, ES, FR, GB, IT, LI, NL, SE, PT				
JP	2002502458	T	20020122	JP 1999-501485	19980602
AT	216407	T	20020515	AT 1998-929411	19980602
ES	2175725	T3	20021116	ES 1998-929411	19980602
BR	9810085	A	20000808	BR 1998-10085	19981210
US	6297314	B1	20011002	US 2000-424092	20000112
PRAI	DE 1997-19723504	A	19970605		
	WO 1998-EP3285	W	19980602		

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

AB Polyurethane coatings with good resistance to weather, chems., moisture, and impact for plastics are manufactured from compns. containing ≥ 1 polyester with OH number 80-200 mg KOH/g and acid number <10 mg KOH/g, ≥ 1 polyacrylate with OH number 80-200 mg KOH/g and acid number <20 mg KOH/g, ≥ 1 di- and(or) polyisocyanate having (un)blocked NCO groups, ≥ 1 UV absorber, ≥ 1 sterically hindered amine having amino ether groups as light stabilizer, and ≥ 1 solvent.

OSC.G 3 THERE ARE 3 CAPLUS RECORDS THAT CITE THIS RECORD (3 CITINGS)

RE.CNT 1 THERE ARE 1 CITED REFERENCES AVAILABLE FOR THIS RECORD

ALL CITATIONS AVAILABLE IN THE RE FORMAT

L3 ANSWER 22 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
AN 1998:639448 CAPLUS
DN 129:343551
OREF 129:69977a,69980a
TI Zirconium-catalyzed and zirconium-promoted cyclization reactions of non-conjugated dienes with alkylmagnesium halides to give cycloalkylmethylmagnesium derivatives
AU Negishi, E.; Rousset, Christophe J.; Choueiry, Daniele; Maye, John P.; Suzuki, Noriyuki; Takahashi, Tamotsu
CS Dep. Chem., Purdue Univ., West Lafayette, IN, 47907, USA
SO Inorganica Chimica Acta (1998), 280(1-2), 8-20
CODEN: ICHAA3; ISSN: 0020-1693
PB Elsevier Science S.A.
DT Journal
LA English
OS CASREACT 129:343551
AB The stoichiometric reaction of certain non-conjugated dienes with n-Bu₂ZrCp₂ provides the corresponding zirconabicycles, such as trans-3-bis(cyclopentadienyl)zirconabicyclo[3.3.0]octane, that can be fully characterized by spectroscopic means. Their treatment with EtMgBr or n-BuMgCl in THF gives the corresponding monocyclic monomagnesium derivs. along with the corresponding alkene-ZrCp₂ derivs. in high yields. In cases where the Grignard reagent is either sterically hindered or lacking β-H, little or no reaction may occur, although some, e.g., s-BuMgCl, react, albeit slowly, to give the expected products in high yields. In cases where either a Grignard reagent in di-Et ether or a dialkylmagnesium (irresp. of solvent) is used, the major product is the corresponding dimagnesio derivative. A couple of intramol. transmetalation paths are proposed for these cases. The Cp₂ZrCl₂-catalyzed reaction of 1,6-heptadiene with EtMgBr fails to induce the desired bicyclization-ring opening sequence to give the corresponding monocyclic monomagnesium derivs. On the other hand, the corresponding reaction with n-BuMgBr does proceed as desired to give the monocyclic monomagnesium product which contains a minor amount of the corresponding exo-methylene derivative. Only traces, if any, of the corresponding dimagnesio derivs. reported to be the major products in Et₂O are formed. This procedure has been applied to catalytically convert several other dienes, i.e., (E)-1-phenyl-1,6-heptadiene, 2,4,4-trimethyl-1,6-heptadiene, diallyl(benzyl)amine, 1,7-octadiene and 1,2-diallylbenzene, into the corresponding monocyclized compds. in moderate to excellent combined yields.
OSC.G 22 THERE ARE 22 CAPLUS RECORDS THAT CITE THIS RECORD (22 CITINGS)
RE.CNT 63 THERE ARE 63 CITED REFERENCES AVAILABLE FOR THIS RECORD
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L3 ANSWER 23 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 1998:151193 CAPLUS
 DN 128:168827
 OREF 128:33257a,33260a
 TI Stable polyorganosiloxane-based compositions with crosslinkable functional groups and their use for producing antiadhesive coating
 IN Priou, Christian; Soldat, Andre; Kerr, Stuart R., III; Beaty, Reeshemah
 PA Rhone-Poulenc Chimie SA, Fr.; Priou, Christian; Soldat, Andre; Kerr, Stuart R., III; Beaty, Reeshemah
 SO PCT Int. Appl., 30 pp.
 CODEN: PIXXD2
 DT Patent
 LA French
 FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI WO 9807798	A1	19980226	WO 1997-FR1492	19970814
W: AL, AU, BA, BB, BG, BR, CA, CN, CU, CZ, EE, GE, HU, IL, IS, JP, KP, KR, LC, LK, LR, LT, LV, MG, MK, MN, MX, NO, NZ, PL, RO, SG, SI, SK, TR, TT, UA, US, UZ, VN, YU, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM				
RW: GH, KE, LS, MW, SD, SZ, UG, ZW, AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE, SN, TD, TG				
FR 2752582	A1	19980227	FR 1996-10330	19960821
FR 2752582	B1	20030613		
CA 2264013	C	19980226	CA 1997-2264013	19970814
CA 2264013	A1	19980226		
AU 9740189	A	19980306	AU 1997-40189	19970814
AU 732617	B2	20010426		
EP 920483	A1	19990609	EP 1997-937629	19970814
EP 920483	B1	20020213		
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, FI, RO				
JP 2000516660	T	20001212	JP 1998-510462	19970814
JP 4190585	B2	20081203		
AT 213263	T	20020215	AT 1997-937629	19970814
ES 2168660	T3	20020616	ES 1997-937629	19970814
PT 920483	E	20020628	PT 1997-937629	19970814
US 6218445	B1	20010417	US 2000-242713	20000110
PRAI FR 1996-10330	A	19960821		
WO 1997-FR1492	W	19970814		

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

OS MARPAT 128:168827
 AB The title radiation-crosslinkable, storage-stable composition is based on ≥1 polyorganosiloxane with crosslinkable epoxy or vinyl ether functional groups, and contains a stabilizing amine agent, and an initiator system for hardening under radiation, in particular UV radiation. The amine is a secondary or tertiary amine, a sterically hindered cyclic amine, or an amine consisting of both types of groups. A composition contained TMS-terminated (epoxycyclohexyl)ethyl Me, di-Me siloxane, ditolyliodinium tetrakis(pentafluorophenyl)borate, and trioctylamine.

OSC.G 8 THERE ARE 8 CAPLUS RECORDS THAT CITE THIS RECORD (9 CITINGS)
 RE.CNT 3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD
 ALL CITATIONS AVAILABLE IN THE RE FORMAT

L3 ANSWER 24 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN

AN 1997:234099 CAPLUS

DN 126:225981

OREF 126:43703a,43706a

TI Preparation of heat-resistant polyurethane-polyurea foam

IN Dempsey, Michael P.

PA Bayer A.-G., USA

SO Can. Pat. Appl., 29 pp.

CODEN: CPXXEB

DT Patent

LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	CA 21/2678	A1	19961118	CA 1996-21/2678	19960326
PRAI US	1995-442624	A	19950517		
AB Title foam is prepared by reacting, at isocyanate index 100-115, (a) an organic polyisocyanate with (b) 40-65 weight%, based on the total amount of components (b), (c), and (d), of an isocyanate-reactive compound containing ≥ 2 isocyanate-reactive groups and having a number average mol. weight of 400-10,000;					
(c) 10-35 weight%, based on total of (b), (c), and (d), of an isocyanate-reactive compound containing ≥ 2 isocyanate-reactive groups other than amino groups and having a mol. weight of 32-399; (d) 5-35 weight%, based on total of (b), (c), and (d), of an organic amine containing ≥ 1 amino group having a moderated reactivity toward the organic polyisocyanate, as indicated by a gel time >.apprx.5 s measured from the material initiation time at 25-40°C, and number average mol. weight 86-400, selected from (1) sterically hindered aromatic amines in which ≥ 1 aromatic ring substituents are ortho to the amino groups, (2) aromatic amines other than amines (d)(1) in which ≥ 1 of the amino groups exhibits reduced reactivity due primarily to electronic effects rather than steric factors, (3) amines having secondary amino groups, (4) nonarom. amines having sterically hindered primary amino groups, and (5) mixts. thereof; and (e) optionally, rigid fibers. The products are useful for automotive applications (no data). Thus, Mondur 489 containing 34.9 parts milled glass fiber/100 parts polyol was reacted with a polyol composition comprising ethylene glycol 18, polyethylene glycol-polypropylene glycol glyceryl ether 55, and polypropylene oxide ether with ethylenediamine 3 parts, containing 20 parts Ethacure 300 to form a plaque having d. 0.5 g/cm ³ , flexural modulus 865 MPa, and parallel heat sag 5.3 mm and perpendicular heat sag 8.5 mm (6-in. overhang, 121°, 1 h), compared with 0.5, 779, 20.5, and 27.5, resp., for a sample without the Ethacure 300.					

L3 ANSWER 25 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 1995:994682 CAPLUS
 DN 124:10548
 OREF 124:2191a,2194a
 TI Elastic amine-modified epoxy resin composition, especially suitable as one-component adhesive
 IN Marten, Manfred; Wehner, Bernhard
 PA Hoechst A.-G., Germany
 SO Eur. Pat. Appl., 18 pp.
 CODEN: EPXXDW
 DT Patent
 LA German
 FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI EP 675185	A2	19951004	EP 1995-104296	19950323
EP 675185	A3	19960110		
R: AT, BE, CH, DE, DK, ES, FR, GB, IE, IT, LI, NL, PT, SE				
DE 4410786	A1	19951005	DE 1994-4410786	19940328
CA 2145589	A1	19950929	CA 1995-2145589	19950327
JP 08034833	A	19960206	JP 1995-68175	19950327

PRAI DE 1994-4410786 A 19940328

AB The title composition contains a reaction product of a polyepoxide, a polyoxyalkylene amine, and, optionally, a polycarboxylic acid, a reaction product of a polyepoxide and a sterically hindered amine, other epoxides, hardeners, and additives and is especially useful as a 1-component adhesive for bonding steel parts in the manufacture of automobiles. A reaction product of a bisphenol A epoxy resin and Jeffamine M 600, a reaction product of polypropylene glycol diglycidyl ether (Beckopox EP 075) and 2-aminobutane, dicyandiamide, and silica were used in an adhesive for bonding steel plates.

L3 ANSWER 26 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
AN 1994:272068 CAPLUS
DN 120:272068
OREF 120:48195a,48198a
TI Chemical strategies for compatibilization of immiscible polymer blends:
polyphenylene ether-nylon copolymer formation through nucleophilic
displacement on aryloxytriazine-capped polyphenylene ethers
AU Brown, S. Bruce
CS Polym. Chem. Lab., Gen. Electr. Res Dev. Cent., Schenectady, NY, 12301,
USA
SO Polymer Preprints (American Chemical Society, Division of Polymer
Chemistry) (1992), 33(2), 598-9
CODEN: ACPPAY; ISSN: 0032-3934
DT Journal
LA English
AB The OH end-groups of poly(phenylene ether) (PPE) were
functionalized by 3 different methods, i.e., chloride displacement from a
chlorodiarloxytriazine, transesterification with a tris(aryloxytriazine)
and branching and/or chain-extension with a dichlorotriazine or with
cyanuric chloride. Coextrusion of the aryloxytriazine-functionalized PPE
with amine-terminated nylons resulted in PPE-nylon copolymer
formation through nucleophilic displacement of an aryloxy group from the
triazine-functionalized PPE by the amine end-groups of the
nylon. The resulting compatibilized blends had excellent mech.
properties. The displacement to form PPE-nylon copolymer competed with
the displacement to form triazine end-capped nylon and no copolymer. The
sterically hindered 2,6-dimethylphenoxy group of PPE was
probably more difficult to displace during melt extrusion than an
unhindered phenoxy group on the triazine ring.
OSC.G 7 THERE ARE 7 CAPLUS RECORDS THAT CITE THIS RECORD (7 CITINGS)

L3 ANSWER 27 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 1994:135387 CAPLUS
 DN 120:135387
 OREF 120:23881a,23884a
 TI Amine-modified metathesis catalysts for production of polymers and copolymers of cycloalkene of norbornene type
 IN Stohandl, Jiri; Vozka, Pavel; Varekova, Irena; Krafiat, Miroslav; Ondruj, Jiri; Mejzlik, Jiri; Balcar, Hynek; Stpanek, Kamil; Heller, Gerhardt; Lederer, Jaromir
 PA Chemopetrol, S.P. Litvinov Vyzkumny Ustav Makromolekularni Chemie, Czech.
 SO PCT Int. Appl., 34 PP.
 CODEN: PIXXD2

DT Patent
 LA English
 FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 9312158	A1	19930624	WO 1992-CS34	19921216
	W: HU, US				
	RW: AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE				
	CZ 279861	B6	19950712	CZ 1991-3808	19911216
	SK 278739	B6	19980204	SK 1991-3808	19911216
	EP 571579	A1	19931201	EP 1992-924538	19921216
	EP 571579	B1	19980401		
	R: BE, DE, FR, GB, IT, NL				
	US 5455318	A	19951003	US 1993-104168	19931015
PRAI	CS 1991-3808	A	19911216		
	WO 1992-CS34	W	19921216		

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

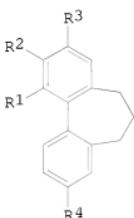
OS MARPAT 120:135387

AB A sterically hindered amine modifies
 ≥1 W compound containing Cl, Br, and/or I to form catalyst precursor which is used with cocatalyst selected from Al, Sn, Pb, Si, Li, Mg, B, Zn, and/or Ge compds., optionally moderator compds., e.g. ethers, in the manufacture of copolymers of norbornene type by reaction injection molding. WCl₆ was modified with 2,2,6,6-tetramethyl-4-hydroxy-4-aminocarbonylpiperidine at mol ratio 1:1 in PhMe and mixed with dicyclopentadiene (I), then admixed with cocatalyst solution at final mol ratio W component:diethylaluminum chloride:di-Bu ether:I at 1:8:20:2000, and heated at 60° to give polymer.

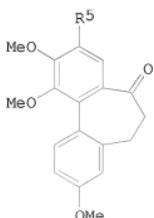
OSC.G 1 THERE ARE 1 CAPLUS RECORDS THAT CITE THIS RECORD (1 CITINGS)
 RE.CNT 2 THERE ARE 2 CITED REFERENCES AVAILABLE FOR THIS RECORD

ALL CITATIONS AVAILABLE IN THE RE FORMAT

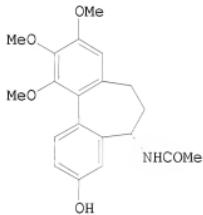
L3 ANSWER 28 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 1993:39215 CAPLUS
 DN 118:39215
 OREF 118:7159a, 7162a
 TI Natural products. Antitubulin effect of congeners of N-acetylcolchinchyl methyl ether: synthesis of optically active 5-acetamidodeaminocolchinchyl methyl ether and of demethoxy analogs of deaminocolchinchyl methyl ether
 AU Boye, O.; Brossi, A.; Yeh, H. J. C.; Hamel, E.; Wegrzynski, B.; Toome, V.
 CS Lab. Struct. Biol., Natl. Inst. Health, Bethesda, MD, 20892, USA
 SO Canadian Journal of Chemistry (1992), 70(5), 1237-49
 CODEN: CJCHAG; ISSN: 0008-4042
 DT Journal
 LA English
 OS CASREACT 118:39215
 GI



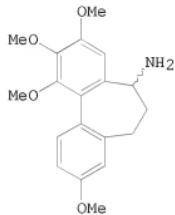
I



II



III



IV

AB Trimethoxy-substituted dihydronaphthalene derivatives I ($R_1 = R_2 = R_4 = \text{MeO}$, $R_3 = \text{H}$; $R_1 = R_3 = R_4 = \text{MeO}$, $R_2 = \text{H}$; $R_2 = R_3 = R_4 = \text{MeO}$, $R_1 = \text{H}$; $R_1 = R_2 = R_3 = \text{MeO}$, $R_4 = \text{H}$), required for a structure-activity study measuring the inhibition of tubulin polymerization *in vitro*, were synthesized by four different routes: (1) Synthesis of I ($R_1 = R_2 = R_4 = \text{MeO}$, $R_3 = \text{H}$) was achieved from 2,3-dimethoxybenzaldehyde via 2,3-(MeO)₂C₆H₃C₆H₃(OMe)CHO-4,2, chain lengthening to propionic acid, acid-catalyzed cyclization toward dihydronaphthalene derivative II ($R_5 = \text{H}$), and removal of the carbonyl group. (2) Compound I ($R_1 = R_3 = R_4 = \text{MeO}$, $R_2 = \text{H}$) was obtained by eliminating the sterically most hindered methoxy group in the oxime of II ($R_5 = \text{MeO}$) by metal reduction in alc. (3) Compound I ($R_2 = R_3 = R_4 = \text{MeO}$,

R1

= H) was prepared from the corresponding biphenyl aldehyde obtained by Grignard reaction on an oxazoline. (4) Compound I (R1 = R2 = R3 = MeO, R4 = H) was obtained by reductive deoxygenation of N-acetylcolchinol (III) tetrazolyl ether derivative. The key role of the aromatic oxygen atoms in colchicine and allo congeners as points of interaction with the colchicine binding site on tubulin was demonstrated by the lack of inhibitory activity of compds. I. Optically active 5-acetamide isomers of N-acetylcolchinyl Me ether were obtained after chemical resolution of amine IV. The absolute configuration of the optical isomers of IV was determined by 1H NMR and CD measurements. These compds. were found inactive as inhibitors of tubulin polymerization

OSC.G 17 THERE ARE 17 CAPLUS RECORDS THAT CITE THIS RECORD (18 CITINGS)

L3 ANSWER 29 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
AN 1992:427623 CAPLUS
DN 117:27623
OREF 117:5014h,5015a
TI The effects of steric hindrance on sub-glass transitions in epoxy polymers
AU Balizer, Edward; Duffy, James V.
CS Nav. Surf. Warf. Cent., Silver Spring, MD, 20903-5000, USA
SO Polymer (1992), 33(10), 2114-22
CODEN: POLMAG; ISSN: 0032-3861
DT Journal
LA English
AB Pairs of sterically hindered and unhindered linear aliphatic and aromatic diamines were prepared and used as curatives for the diglycidyl ether of bisphenol A. The steric hindrance was caused by Me group substitution of a H atom adjacent to the amine . For each pair, the hindered diamine cure had a lower d. and a higher glass transition. Another pair of diamines was prepared for which the Me group was replaced by Et and Bu side chains; for these resins, both the d. and glass transition decreased. Torsional pendulum results showed that the sub-glass transition for the hindered cures shifted to lower temps. and had a greater activation energy. Anal. by the Havriliak-Negami dispersion equation showed that the hindered resins had broader and more sym. relaxations. The background hysteresis loss outside of the relaxation region was analyzed by the Nutting equation and was found to decrease with steric hindrance.

L3 ANSWER 30 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 1992:7559 CAPLUS
 DN 116:7559
 OREF 116:1467a,1470a
 TI Manufacture of polyurethane-polyurea and/or polyurea reaction-injection moldings with improved coatability
 IN Ueda, Hiroshi; Uchida, Yuji; Yoshida, Yoshio; Inoue, Hiroshi; Kaneda, Toshikazu; Moriya, Toshiaki; Kumazawa, Tsutomu
 PA Mitsui Toatsu Chemicals, Inc., Japan
 SO Jpn. Kokai Tokkyo Koho, 8 pp.
 CODEN: JKXXAF
 DT Patent
 LA Japanese
 FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI JP 03195770	A	19910827	JP 1989-332643	19891225
PRAI JP 1989-332643				
OS MARPAT 116:7559				

AB Title moldings are prepared from polyisocyanate, polyisocyanate-reactive polymer (A) having mol. weight 800-12,000, 5-50% (based on A) sterically hindered aromatic diamine as chain extender, and 0.01-2.0% (based on A) sterically hindered amine-type light stabilizer and possibly other conventional stabilizers. Thus, reaction injection molding a formulation of ethylene oxide-propylene oxide copolymer glycerin ether (OH number 34, containing 85% propylene oxide, having 15% ethylene oxide at ends) 100, an 80:20 mixture of 1-methyl-3,5-diethyl-2,4-diaminobenzene and 1-methyl-3,5-diethyl-2,6-diaminobenzene as chain extenders 25, Dabco 33LV 0.15, DBTDL 0.15, Sanole LS 770 as light stabilizer 0.5, and tripropylene glycol-MDI prepolymer (NCO index 100) 64 g and postcuring 30 min at 120° gave moldings with excellent resistance to discoloration and yellowing.

OSC.G 1 THERE ARE 1 CAPLUS RECORDS THAT CITE THIS RECORD (1 CITINGS)

L3 ANSWER 31 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 1991:453266 CAPLUS
 DN 115:53266
 OREF 115:9201a,9204a
 TI Antioxidant for lubricants and lubricant compositions
 IN Evans, Samuel
 PA Ciba-Geigy A.-G., Switz.
 SO Eur. Pat. Appl., 38 pp.
 CODEN: EPXXDW

DT Patent
 LA German

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	EP 406826	A1	19910109	EP 1990-112762	19900704
	EP 406826	B1	19930811		
	R: BE, DE, DK, ES, FR, GB, IT, NL				
	CA 2020558	A1	19910108	CA 1990-2020558	19900705
	CA 2020558	C	20011106		
	BR 9003187	A	19910827	BR 1990-3187	19900705
	DD 297443	A5	19920109	DD 1990-342534	19900705
	JP 03045696	A	19910227	JP 1990-179385	19900706
	JP 2943004	B2	19990830		
	ZA 9005305	A	19910529	ZA 1990-5305	19900706
	CN 1048560	A	19910116	CN 1990-103183	19900707
	CN 1028243	C	19950419		
	KR 151400	B1	19981001	KR 1990-10358	19900707
	US 5268113	A	19931207	US 1992-933599	19920820
PRAI	CH 1989-2528	A	19890707		
	US 1990-546277	B1	19900628		

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

AB The antioxidant contains a sterically hindered amine and a phenol substituted in one or both ortho positions by C1-24 alkyl, C5-12 cycloalkyl, C7-9 phenylakyl, or a CH2SR1 group, where R1 is specific alkyl, ether chain or Ph, and substituted in the para position by H, C1-18 alkyl, or specific S-, O-, or N-containing chains. The additive is especially suitable for motor oils; it reduces sludge formation significantly.

L3 ANSWER 32 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 1990:159257 CAPLUS
 DN 112:159257
 OREF 112:26935a,26938a
 TI Manufacture of polyoxyalkylene alkylamine derivatives
 IN Gerkin, Richard Michael; Kirchner, David Lee
 PA Union Carbide Corp., USA
 SO Eur. Pat. Appl., 7 pp.
 CODEN: EPXXDW

DT Patent
 LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	EP 343486	A2	19891129	EP 1989-108795	19890516
	EP 343486	A3	19900114		
	EP 343486	B1	19940629		
	R: AT, BE, CH, DE, ES, FR, GB, GR, IT, LI, LU, NL, SE				
	US 4960942	A	19901002	US 1988-195010	19880517
	JP 02064126	A	19900305	JP 1989-120623	19890516
	JP 06049754	B	19940629		
	CA 1339907	C	19980609	CA 1989-599756	19890516
PRAI US	1988-195010	A	19880517		

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

AB The title compds. are prepared from polyethers bearing NH₂ groups by catalytic alkylation with sterically hindered alcs. in the presence of H at 175-250°/250-2000 psi. Heating Jeffamine T-5000 [trimethylolpropane polypropylene glycol ether (1:3) tris(aminopropyl) ether] 1204.1, iso-PrOH 424.7, and Ni catalyst 42.1 g with H at 190°/200 psi for 20 h gave a product containing 86% secondary amine; vs. 4% with MeOH instead of iso-PrOH.

OSC.G 3 THERE ARE 3 CAPLUS RECORDS THAT CITE THIS RECORD (3 CITINGS)

L3 ANSWER 33 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 1988:151730 CAPLUS
 DN 108:151730
 OREF 108:24921a,24924a
 TI Recycling of rigid polyurethane wastes
 IN Ionescu, Mihail; Dumitriu, Viorica Tatiana; Mihalache, Ioana; Stoenescu,
 Felicia; Mihai, Stancă
 PA Combinatul Petrochimic Midia, Navodari, Rom.
 SO Rom., 4 pp.

CODEN: RUXXA3
 DT Patent
 LA Romanian
 FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI RO 89944	B1	19860830	RO 1984-114849	19840613
PRAI RO 1984-114849		19840613		

AB Rigid polyurethane foam wastes are simultaneously aminolyzed by NH₃, ethylene diamine (I), diethylene triamine, hexamethylenediamine, or ethanolamine and alkoxylated by ethylene oxide, propylene oxide (II), butylene oxide, Ph glycidyl ether, or styrene oxide optionally in the presence of a HO group-containing, sterically hindered tert-amine at aminolyzing amine-waste ratio 1:(1-2) and 160-200° to rapidly give polyols, useful in the manufacture of polyurethane foams. Thus, adding 270 mL II in 1.5 h to a mixture containing 100 g I, 100 g crumbled rigid polyurethane foam waste, and 2 mL dimethylethanolamine catalyst at 160-190° and 4 kg/cm² N pressure, and reaction mixture was stirred an addnl. 1 h at 110-115°/1.2 kg/cm² to give a brown liquid polyol with OH index 522 mg KOH/g, acid index 1.8 mg KOH/g, and viscosity 5220 cP, from which a polyurethane foam with satisfactory properties could be prepared

OSC.G 1 THERE ARE 1 CAPLUS RECORDS THAT CITE THIS RECORD (1 CITINGS)

L3 ANSWER 34 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 1987:125716 CAPLUS
 DN 106:125716
 OREF 106:20437a,20440a
 TI Cosmetics containing amide-amine condensates for protecting hair and skin against light
 IN Mahieu, Claude; Papantoniou, Christos
 PA Oreal S. A. , Fr.
 SO Ger. Offen., 13 pp.
 CODEN: GWXXBX

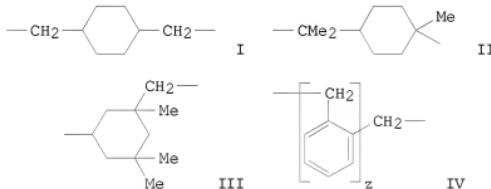
DT Patent

LA German

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	DE 3622066	A1	19870108	DE 1986-3622066	19860701
	FR 2584292	A1	19870109	FR 1985-10158	19850703
	FR 2584292	B1	19871016		
	FR 2600249	A2	19871224	FR 1986-8858	19860619
	FR 2600249	B2	19880916		
	BE 905033	A1	19870102	BE 1986-216861	19860702
	NL 8601723	A	19870202	NL 1986-1723	19860702
	GB 2177916	A	19870204	GB 1986-16124	19860702
	GB 2177916	B	19890726		
	US 4844889	A	19890704	US 1986-881279	19860702
	CA 1279262	C	19910122	CA 1986-512929	19860702
	JP 62026211	A	19870204	JP 1986-155223	19860703
PRAI	FR 1985-10158	A	19850703		
	FR 1986-8858	A	19860619		

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT
 GI



AB The title cosmetics contain a saturated condensate obtained by the addition of
 a bisacrylamide H2C:CHCONHCHR1NHCOCH:CH2 (R1 = H, C1-4 alkyl) to ≥1 sterically-hindered diamine H2NRNH2 [R = (CH2)nCR2Me(CH2)m, I, II, III, IV, etc.; R2 = H, Me; n,m = 0, 1, z = 1, 2] followed by saturation (hydrogenation or thiol or amine addition). A saturated polycondensate was prepared by the polyaddn. of (3-aminomethyl-3,5-trimethyl)cyclohexylamine with methylenebisacrylamide, followed by saturation with cysteine. A shampoo contained the condensate 1 g, Na ethoxylated C12-14 alkyl ether sulfate 5g, 30% cocoamido Pr betaine 3 g, HCl to pH 7, and water to 100 g.

10/591,778

The condensate increased the mech. strength of bleached human hair.
OSC.G 1 THERE ARE 1 CAPLUS RECORDS THAT CITE THIS RECORD (2 CITINGS)

L3 ANSWER 35 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
AN 1987:49589 CAPLUS
DN 106:49589
OREF 106:8203a,8206a
TI Preparation of hindered lithium amide bases and rates of their reaction with ether solvents
AU Kopka, Ihor E.; Fataftah, Zacharia A.; Rathke, Michael W.
CS Dep. Chem., Michigan State Univ., East Lansing, MI, 48824, USA
SO Journal of Organic Chemistry (1987), 52(3), 448-50
CODEN: JOCEAH; ISSN: 0022-3263
DT Journal
LA English
OS CASREACT 106:49589
AB The conversion of hindered secondary amines EtCR1R2NHCR3R4Et [R1, R2 = Me, Et; R1R2 = R3R4 = (CH₂)₅] to the corresponding lithium amides was examined under a variety of conditions. The more hindered secondary amines were inert to MeLi, sec-butyllithium, and BuLi in the absence of solvent additives. In the presence of Me₂NCH₂CH₂NMe₂, all of the amines reacted with BuLi at reasonable rates (<24 h, 25°). The lithium amides reacted fairly rapidly with THF solvent (T_{1/2} < 10 h), with the exception of LiN(CHMe₂)₂. Reactions with Et₂O were somewhat slower.
OSC.G 14 THERE ARE 14 CAPLUS RECORDS THAT CITE THIS RECORD (14 CITINGS)

L3 ANSWER 36 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
AN 1986:167908 CAPLUS
DN 104:167908
OREF 104:26579a,26582a
TI Low-temperature EPR and near-infrared MCD studies of highly anisotropic low-spin ferriheme complexes
AU Gadsby, Paul M. A.; Thomson, Andrew J.
CS Sch. Chem. Sci., Univ. East Anglia, Norwich, NR4 7TJ, UK
SO FEBS Letters (1986), 197(1-2), 253-7
CODEN: FEBLAL; ISSN: 0014-5793
DT Journal
LA English
AB Low-temperature ESR spectra and near-IR MCD spectra of the bis complexes of imidazole, 1-methylimidazole, 4-methylimidazole, 1,2-dimethylimidazole, and 2-methylimidazole with Fe(III) octaethylporphyrin (OEP) in the mixed organic solvent dichloromethane/diethyl ether are reported. The latter 2 complexes have highly anisotropic ESR spectra characteristic of the low-spin Fe(III) state. The optical charge-transfer bands have an unusually high MCD intensity, with a narrow linewidth. This feature may be typical of bis-histidine-ligated heme in cytochromes with sterically strained coordination and may be used to diagnostic such a conformation. The ESR and near-IR MCD spectra of the bis-butylamine complex of Fe(III) OEP are also reported. Bis-amine-ligated heme may be clearly distinguished from sterically hindered bis-imidazole (bis-histidine) by near-IR MCD spectroscopy, whereas the assignment is uncertain using ESR spectroscopy alone.
OSC.G 5 THERE ARE 5 CAPLUS RECORDS THAT CITE THIS RECORD (5 CITINGS)

L3 ANSWER 37 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 1985:148717 CAPLUS
 DN 102:148717
 OREF 102:23353a,23356a
 TI Secondary aminoether alcohols
 IN Stogryn, Eugene L.; Ho, W. S. Winston; Montagna, Angelo A.; Sartori, Guido
 PA Exxon Research and Engineering Co. , USA
 SO U.S., 9 pp.
 CODEN: USXXAM

DT Patent
 LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 4487967	A	19841211	US 1983-565097	19831223
	CA 1228368	A1	19871020	CA 1984-465354	19841012
	NO 8404388	A	19850624	NO 1984-4388	19841105
	NO 160362	B	19890102		
	NO 160362	C	19890412		
	EP 147990	A1	19850710	EP 1984-308791	19841217
	EP 147990	B1	19880622		
	R: BE, CH, DE, FR, GB, IT, LI, NL				
	AU 8437096	A	19850704	AU 1984-37096	19841221
	AU 568708	B2	19880107		
	JP 60169450	A	19850902	JP 1984-270550	19841221
	JP 04081978	B	19921225		
	BR 8406684	A	19851022	BR 1984-6684	19841221
PRAI	US 1983-565097	A	19831223		

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

AB A severely sterically hindered secondary amino ether alc. was prepared from a primary amine and a polyalkenyl ether glycol over a supported hydrogenation catalyst at elevated temps. Thus, Ni-Cu-Co/SiO₂/kieselguhr (Trimmelitic 2330) activated with H₂ at 350° catalyzed the reaction of Me₃CNH₂ with (HOCH₂CH₂)₂₀ (2:1) to give 40% Me₃CNH(CH₂)₂₀(CH₂)₂₀H (I) and byproduct N-tert-butylmorpholine (II) (I/II = 4.3). Also used as catalyst in this reaction were supported Ni, supported Rh, etc.

OSC.G 10 THERE ARE 10 CAPLUS RECORDS THAT CITE THIS RECORD (10 CITINGS)
 RE.CNT 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD
 ALL CITATIONS AVAILABLE IN THE RE FORMAT

L3 ANSWER 38 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
AN 1983:73244 CAPLUS
DN 98:73244
OREF 98:11227a,11230a
TI Kinetics of a sterically hindered amine-cured epoxy resin system
AU Buckley, Leonard J.; Roylance, David K.
CS Nav. Air Dev. Cent., Warminster, PA, USA
SO SAMPE Quarterly (1982), 14(1), 8-13
CODEN: SAMQA2; ISSN: 0036-0821
DT Journal
LA English
AB Fourier-transform IR spectroscopy (FT-IR) and torsional braid anal. (TBA) were used to study the reaction of an epoxy resin system cured with a sterically hindered amine. The system consisted of a 2/3:1/3 stoichiometric mixture of DER 332 [25085-99-8] and o-cresol-formaldehyde novolak resin polyglycidyl ether (ECN 1299) hardened with 2,5-dimethyl-2,5-hexanediamine [23578-35-0]. TBA showed the apparent activation energy to be .apprx.42 kJ/mol. The reaction kinetics was also considered from the unreacted freshly mixed condition and a partially reacted B-staged condition using IR spectroscopy. The B-staged condition was unreactive and stable at room temperature due to the quenching of the primary amine reaction by the glassy structure and the steric hindrance of the secondary amine reaction. Apparent activation energies for these 2 conditions were 4.94 kJ/mol and 49.0 kJ/mol, resp. The storage life at room temperature of the B-staged resin system was predicted to be ≥ 3 mo based on extrapolation of the exptl. kinetic data.

L3 ANSWER 39 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 1981:572712 CAPLUS
 DN 95:172712
 OREF 95:28841a,28844a
 TI Compositions for stabilizing electrolytes in lithium/titanium disulfide systems
 IN Rao, Bhaskara M. L.; Eustace, Daniel J.; Farcasiu, Dan
 PA Exxon Research and Engineering Co. , USA
 SO U.S., 6 pp.
 CODEN: USXXAM
 DT Patent
 LA English
 FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI US 4284692	A	19810818	US 1980-144680	19800428
PRAI US 1980-144680		19800428		

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

AB A Li-TiS₂ battery has an electrolyte of ≥1 Li salt (LiAsF₆) a cyclic ether (dioxolane [646-06-0]-MeOCH₂CH₂OMe) solvent, and an inhibitor of the polymerization of the cyclic ether by TiS₂. The inhibitor is selected from Li_xTiS₂ (x .apprx.0.025-0.1) and/or ≥1 sterically hindered amine which does not intercalate with TiS₂. The possible amines are: 1,8-bis(dimethylamino)naphthalene [20734-58-1]; 2,2,6,6-tetramethylpiperidine; a polymeric pyridine such as poly(vinylpyridine) or 2-vinylpyridine-styrene copolymers [24980-54-9]; or tetramethylenediamine.

OSC.G 4 THERE ARE 4 CAPLUS RECORDS THAT CITE THIS RECORD (4 CITINGS)

L3 ANSWER 40 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
AN 1981:498884 CAPLUS
DN 95:98884
OREF 95:16627a,16630a
TI Stabilization of polyether polyols and polyurethane foams prepared
therefrom
IN Hinze, Kenneth J.
PA Dow Chemical Co., USA
SO U.S., 5 pp.
CODEN: USXXAM
DT Patent
LA English
FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI US 4275173	A	19810623	US 1980-141854	19800421
PRAI US 1980-141854		19800421		

ASSIGNMENT HISTORY FOR US PATENT AVAILABLE IN LSUS DISPLAY FORMAT

AB Polyurethane foams are prepared from polyols stabilized against oxidative degradation with a synergistic combination of sterically hindered phenols, 4,4'-bis(α,α' -dimethylbenzyl)diphenylamine (I) [10081-67-1], and phenothiazine (II) [92-84-2]. Thus, a scorch-resistant polyurethane [57516-88-8] foam was prepared from a polyethylenepolypropylene glycol glycerol ether [9082-00-2] (containing 2,6-di-tert-butyl-4-methylphenol [128-37-0] 1900, I 1000 and II 100 ppm) 200, water 10.4, L-540 silicone surfactant 2.4, Niax A-6 amine catalyst 0.25, T-9 stannous octoate 0.4 and TDI 141.9 g.

OSC.G 4 THERE ARE 4 CAPLUS RECORDS THAT CITE THIS RECORD (4 CITINGS)

L3 ANSWER 41 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN

AN 1980:533555 CAPLUS

DN 93:133555

OREF 93:21303a,21306a

TI Polyethylene compositions for electric insulators

PA Dainichi Nippon Cables, Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 3 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 55065245	A	19800516	JP 1978-138996	19781110
PRAI	JP 1978-138996	A	19781110		
AB Polyethylene (I) [9002-88-4] elec. insulator compns. contain sterically hindered phenol (and/or S-containing ether) and amine antioxidants. For example, a dicumyl peroxide-cured I specimen containing 0.3 phr 4,4'-thiobis(6-tert-butyl-3-methylphenol) (II) [96-69-5] and 0.2 phr 4,4'-bis(α,α -dimethylbenzyl)diphenylamine (III) [10081-67-1] had better treeing resistance (90° water) than a control containing 0.5 phr II and no III.					

L3 ANSWER 42 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
AN 1977:189100 CAPLUS
DN 86:189100
OREF 86:29653a,29656a
TI Oxonium salt alkylation of structurally and optically labile alcohols
AU Diem, Mary Jo; Burow, D. F.; Fry, James L.
CS Bowman-Oddy Lab., Univ. Toledo, Toledo, OH, USA
SO Journal of Organic Chemistry (1977), 42(10), 1801-2
CODEN: JOCEAH; ISSN: 0022-3263
DT Journal
LA English
AB The acid-sensitive alcs. Me₃CCH₂OH, (R)-(+)-PhCHMeOH, and
(S)-(-)-EtCHMeCH₂OH were converted to their Et ethers by treatment with
Et₃O⁺BF₄⁻ in CH₂Cl₂ in the presence of (iso-Pr)₂NET (I). O-methylation
using Me₃O⁺BF₄⁻ was carried out in the presence of the less nucleophilic
base 1,8-bis(dimethylamino)naphthalene, since I was rapidly N-methylated.
Treatment of 2-tert-butyl-2-adamantanol with the above oxonium salt-
sterically hindered amine reagents did not
give the Me or Et ether.
OSC.G 23 THERE ARE 23 CAPLUS RECORDS THAT CITE THIS RECORD (23 CITINGS)

L3 ANSWER 43 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN

AN 1967:464543 CAPLUS

DN 67:64543

OREF 67:12179a,12182a

TI Preparation of aniline chromium tricarbonyl compounds

IN Whiting, Mark C.

PA Ethyl Corp.

SO U.S., 5 pp.

CODEN: USXXAM

DT Patent

LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 3317522		19670502	US 1963-264025	19630311
PRAI	GB		19620312		

GI For diagram(s), see printed CA Issue.

AB Aniline Cr tricarbonyl compds. are synthesized by substitution of F in Ia, where X is H or alkyl, by the corresponding amine R1R2NH (R1 and R2 are H, alkyl or aryl groups). Slow reactions with sterically hindered amines can be accelerated with catalytic quantities of a non-alc. polar solvent (acetamide, formamide, dimethylformamide, acetonitrile, and Me2SO). Thus, HMeNPhCr(CO)3 (I) is obtained in 84% yield by treating 92.7 g. FPhCr(CO)3 and 270 g. MeNH2 for 12 min. at 21°. I m. 123.5° (petroleum ether). Similarly prepared were Ia (X = H) (amine used, % yield, and m.p. given: Me2NH, 84, 144.5-45°; hexylamine, 80, 67-8°; cyclohexylamine, 82, 132-3°; isopropylamine, 85, 102-3°; piperidine, 94, 125-6.5°; pyrrolidine, 92, 161-2°; morpholine, 85, 171-2.5°; benzylamine, 87, 129-30°; Et2NH, 70, 113-15°; EtNH2, 11, 112-16°; tert-BuNH2, 88, 126-8°. Ia are useful as petroleum additives, chemical intermediates, fungicides, herbicides, bactericides, and pesticides.

OSC.G 1 THERE ARE 1 CAPLUS RECORDS THAT CITE THIS RECORD (1 CITINGS)

L3 ANSWER 44 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 1965:15204 CAPLUS
 DN 62:15204
 OREF 62:2738g-h,2739a-c
 TI Mono and diaryl phosphites
 IN Shepard, Alvin F.; Danneis, Bobby F.; Kujawa, Francis M.
 PA Hooker Chemical Corp.
 SO 20 pp.
 DT Patent
 LA French
 FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI FR 1366579	-----	19640710	FR 1963-945453	19620831
BE 651633	-----		BE	
NL 297350	-----		NL	
US 3271481	-----	19660906	US 1962-220902	19620831
PRA FR	-----	19620831		

AB The title compds. were prepared from sterically hindered phenols and PC13 with removal of the HCl formed. The products could be used as stabilizers, plasticizers, fire-retardants, or lubricant additives. To a mixture of 463 g. PC13 and 121.4 g. NET3, 265 g. of 2,4,6-tri-tert-butylphenol was added slowly at 10° under a N atmospheric, the mixture refluxed 6 hrs., Et3N.HCl filtered off and washed with petr. ether, and the combined filtrates distilled to give a fraction, b0.5 119-24°. On chilling, 313 g. of a white crystalline solid was obtained. After grinding, this 2,4,6-tri-tert-butylphenyl dichlorophosphite was kept 2 hrs. with 4 l. H2O, filtered off, dried, and recrystd. (dimethoxyethane), m. 185°. In one preparation, the Na salt of 2,6-di-tert-butylphenol was used in the absence of a tertiary amine. Also, CaO or a vacuum were used in similar prepsn. Also prepared were 2,6-di-tert-butylphenyl dichlorophosphite, pale yellow, b1 143-6°; 2,6-di-tert-butyl-4-methylphenyl dichlorophosphite, b0.5 117-20° (solidified on standing); 2,6-di-tert-butyl-4-nonylphenyl dichlorophosphite, b2.5 148-154°; 2,6-di-tert-butyl-4-chlorophenyl dichlorophosphite, b2.6 156-66°; 2,4,6-tris-(phenethyl)phenyl dichlorophosphite; mono(2,6-di-tert-butylphenyl) phosphite, m. 136.5-38° (hexane); mono(2,6-di-tert-butyl-4-methylphenyl) phosphite, m. 190-5°; mono(2,6-di-tert-butyl-4-nonylphenyl) phosphite, a viscous oil (Na salt had good surface active properties); mono(2,6-di-tert-butyl-4-chlorophenyl) phosphite, m. 140-2° (dimethoxyethane); and mono[2,4,6-tris(phenethyl)phenyl] phosphite, a viscous oil. Diaryl chlorophosphites were prepared as follows. A mixture of 112 g. di-tert-butyl-4-methylphenol and 117 g. NET3 was heated to 75°. After a clear solution formed, 20.5 g. PC13 was added over 1.5 hrs., the mixture heated 19 hrs. at 111-13°, cooled, 156 g. petr. ether added, the solids washed with petr. ether, the combined filtrates distilled in vacuo to eliminate excess phenol as well as solvent, the residue (58 g.) ground and treated with 11.5% aqueous NaOH, and the solid filtered off, washed with H2O, and recrystd. from Me2CO to give 2,6-di-tert-butyl-4-methylphenyl monochlorophosphite, m. 113-15°. This (10 g.) was hydrolyzed in 500 cc. 0.1N HCl by refluxing 24 hrs. in an inert atmospheric. After filtration, washing free of Cl-, and drying, the bis(2,6-di-tert-butyl-4-methylphenyl) phosphite m. 161-3°. Recrystn. from hexane did not alter the m.p. appreciably. Also prepared were bis(2,6-di-tert-butyl-4-chlorophenyl) chlorophosphite, b2.6 156-66°; bis(2,4,6-tri-tert-butylphenyl) chlorophosphite, m. 173-4° (hexane); bis(2,6-di-tert-butylphenyl) chlorophosphite; bis(2,6-di-tert-butyl-4-chlorophenyl phosphite, m. 147-8.5°; bis(2,6-di-tert-butylphenyl) phosphite, m. 147-9° (Me2CO-hexane);

and (2,6-di-tert-butylphenyl) diphenyl phosphite, b. 200-2°. The products were quite stable to hydrolysis.

L3 ANSWER 45 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 1964:45745 CAPLUS
 DN 60:45745
 OREF 60:8035f-h,8036a-d
 TI Synthesis and cyclodimerization of stable
 1-tert-alkylamino-2,3-epoxypopropanes-related sterically promoted
 eight-membered ring closures
 AU Gaertner, V. R.
 CS Monsanto Chem. Co., St. Louis, MO
 SO Tetrahedron Letters (1964), (3-4), 141-6
 CODEN: TELEAY; ISSN: 0040-4039
 DT Journal
 LA Unavailable
 OS CASREACT 60:45745
 GI For diagram(s), see printed CA Issue.
 AB Exceptionally stable secondary amino epoxides, their cyclodimerization to diazacyclooctanediois, and novel 8-membered ring closures are described. Equimolar reaction of tert-alkylamines with epichlorohydrin (I), either reactant being present in 25-100% excess in MeOH at 20-5° followed by dehydrohalogenation with 40-50% aqueous alkali below 35° of the crude 1-tert-alkylamino-3-chloro-2-propanols gave the following 1-tert-alkylamino-2,3-epoxypopropanes (II) (alkyl group, b.p./mm., n_{25D}, and % yield given): tert-Bu (III), 54-5°/10, 1,4307, 58; tert-BuCH₂CMe₂ (IV), 65-6°/1, 1,4492, 66. When atmospheric moisture was excluded, pure III and IV could be kept at room temperature for some time without measurable change. H₂O or acls. catalyzed the disappearance of the oxirane ring. In MeOH at 25.0°, the 2nd order constant for the initial dimerization of III was 7.0 ± 10^{-6} l. mole⁻¹ sec.⁻¹. This rate is only 3.5-fold slower than the initial reaction of the less hindered tert-BuNH₂ with I ($k_2 = 2.5 \pm 10^{-5}$ l. mole⁻¹ sec.⁻¹ at 25.0° in 96% MeOH). The stability of II was attributable more to the absence of H-bonding catalysts than to the steric effect. After 2 months in MeOH at 20-5°, less than 10% of the original weight of III was not volatile up to 220°/1 mm. Crystallization of the distillate from heptane and then from petr. ether gave 2 major products, V (R = R₁ = tert-Bu) (VI), one (23% yield) m. 124-5°, and the other m. 69-72° (incompletely separated from unidentified oils). Three alternate syntheses of V were mentioned. The different behaviors of the OH nuclear magnetic resonance peaks upon dilution suggested that the higher and lower melting diols were trans and cis isomers, resp. Another example of cyclodimerization of 1-amino-2,3-epoxypopropanes indicated that the reaction was not limited to highly hindered starting materials. 1-Anilino-2,3-epoxyp propane (VII) gave 17% V (R = R₁ = Ph) (VIII), m. 212-13% also obtained (25% yield) from PhNH₂ and N,N-bis(2,3-epoxypropyl)aniline (IX). An equimolar mixture of III and VII gave VI, VIII, and a trace V (R = tert-Bu, R₁ = Ph) (X), m. 158-9°. The reaction of primary amine with a diglycidylamine was also general and provided V carrying 2 different substituents. Either PhNH₂ and tert-butylbis(2,3-epoxypropyl)amine (XI) or tert-BuNH₂ and IX gave X in 28 and 24% yields, resp. From the 2 appropriate pairs was similarly prepared V (R = Bu, R₁ = Ph), m. 133-4°. In these cases only 1 pure crystalline compound, probably the trans isomer, was isolated. A similar 8-membered ring closure, involving an intermediate mercaptide, occurred in the rapid exothermic reaction of Na₂S with XI to give 27% 1-tert-butyl-1-aza-5-thia-3,7-cyclooctanediol, m. 94.5-5.0°. The structure was established by hydrogenolysis to tert-BuN(CH₂CHMeOH)₂ (XII) (isolated by vapor phase chromatography), identical (infrared spectrum) with XII prepared from tert-BuNH₂ with excess propylene oxide (63% yield, b_{0.6} 85°, n_{25D} 1.4558). These cyclizations are rationalized by a sterically favored conformer of solvated intermediate of type XIII

10/591,778

(A = NHR or S-) which both promotes ring formation and inhibits
polymerization.

OSC.G 2 THERE ARE 2 CAPLUS RECORDS THAT CITE THIS RECORD (2 CITINGS)

L3 ANSWER 46 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 1961:143790 CAPLUS
 DN 55:143790
 OREF 55:27155e-i,27156a-i
 TI New class of local anesthetics. Hydroxyalkyliminobisacetamides
 AU Freed, Meier E.; Bruce, William F.; Hanslick, Roy S.; Maschitti, Albert
 CS Wyeth Labs., Philadelphia, PA
 SO Journal of Organic Chemistry (1961), 26, 2378-83
 CODEN: JOCEAH; ISSN: 0022-3263
 DT Journal
 LA Unavailable
 AB cf. CA 53, 6088e. A series of hydroxyalkyliminobisacetamides, HOXN(CH₂CONR1)CH₂CONR2R₃ (where X is alkylene, cycloalkylene, or aralkylene, R, R₁, R₂, R₃ represent lower alkyl or aralkyl, and where RR1 may or may not equal R2R3), were prepared, examined for local anesthetic action, and studied for structure-activity relationships. The preparation of all chloroacetamides, hydroxyalkylaminooacetamides, hydroxyalkyliminooacetamides and their esters were carried out essentially in the same manner. PhCH₂CMe₂NHMe (0.86 mole) in 500 ml. PhMe stirred 1 hr. at -15° with addition of 0.40 mole ClCH₂COCl, the mixture filtered at 20°, the amine HCl salt washed with PhMe, the combined filtrate and washings dried, and the residue on evaporation distilled yielded 70.5% PhCH₂CMe₂NMeCOCH₂Cl (I), b_{0.5} 140-1°. HOCH₂CH₂NH₂ (0.1 mole) and 30 g. anhydrous powdered Na₂CO₃ in 300 ml. well-stirred boiling BuOH slowly treated with 0.1 mole I in 50 ml. BuOH, the mixture refluxed 12 hrs., cooled, and filtered, and the residue on evaporation crystallized from C₆H₁₄ yielded
 63% HOCH₂CH₂NHCH₂CONMeCMe₂CH₂Ph, m. 74.5-6.5°; HCl salt, m. 163-4°. Similarly were prepared and tabulated hydroxyalkylaminooacetamides, RNHCH₂CONR1Me (R, R₁, and m.p. HCl salt given): PhCHOCH₂, PhCH₂CMe₂, 201-2°; PhCHOCHMe₂, PhCH₂CMe₂, 189-90°; HOCH₂CMe₂, PhCH₂CMe₂, 169-70°; (HOCH₂)₃C, PhCH₂CMe₂, 175-6°; HOCHMeCH₂, PhCH₂, 134-5°. I (0.1 mole) and 20 g. K₂CO₃ in 250 ml. boiling BuOH stirred with addition of 0.05 mole freshly distilled HOCH₂CH₂NH₂, the mixture refluxed 20 hrs. and the cooled mixture filtered, the filtrate washed (aqueous 5% Na₂CO₃, H₂O) and the dried (MgSO₄) solution evaporated in vacuo yielded 71% hydroxyalkyliminobisacetamide, RN(CH₂CONR1R2)₂ (II) (R = HOCH₂CH₂, R₁ = Me, R₂ = PhCH₂CMe₂) (III), m. 104-5°; HCl salt m. 146-7° (MeOH-Me₂CO); nicotinic acid ester m. 158-9°. III (20 g.) in 100 ml. dry CHCl₃ treated with 5 g. SOCl₂ in 25 ml. CHCl₃, the mixture stirred 3 hrs., and the residue on evaporation crystallized from alc.-Et₂O yielded 79 g. II (R = ClCH₂CH₂, R₁ = Me, R₂ = PhCH₂CMe₂) (IV) (3 g.) in 20 ml. MeOH containing 3 g. anhydrous NH₃ heated 18 hrs. at 90° in a pressure tube, the cooled mixture and MeOH rinsings filtered from NH₄Cl, freed from MeOH and excess NH₃, and taken up in 50 ml. Me₂CHOH, and the filtered solution treated with dry HCl and diluted with 150 ml. dry Et₂O yielded 40.5% II (R = H₂NCH₂CH₂, R₁ = Me, R₂ = PhCH₂CMe₂), m. 231-2°. III (0.02 mole) in 150 ml. dry Et₂O added slowly with stirring to 1.8 g. LiAlH₄ in 300 ml. dry Et₂O, the mixture refluxed 25 hrs. before cautious decomposition with 8 ml. H₂O, the dried Et₂O layer treated with HCl, the oily product triturated with Me₂CO, and the product (29.3%) recrystd. from MeOHMe₂CO yielded HOCH₂CH₂N(CH₂CH₂NMeCMe₂CH₂Ph)₂, m. 229-30° (decomposition); tri-HCl salt m. 239-40°; MeI salt, m. 122-3°; tri-MeI salt, m. 154-5°. To obtain the bis compds. with sterically hindered amino alcs., the use of a higher boiling solvent (such as PhOMe) was necessary. Phys. and pharmacol. data are tabulated for the various series of compds., RN(CH₂CONR1R2)₂ (R, R₁, R₂, b.p./mm., duration

of activity on rabbit cornea and % solution given): HOCH₂CH₂, (R1R2=)CH₂CH₂, 203-5°/1.0, neg., 0.1; HOCH₂CH₂, Me(CH₂)₃, Me(CH₂)₃, 208-10°/0.5, 25 min., 0.01; MeCHOHCH₂, Me(CH₂)₃, Me(CH₂)₃, 200-5°/0.1, neg., 0.1; HOCH₂CH₂, Me₂CHCH₂, Me₂CHCH₂, 170-1°/0.5, neg., 0.1; HO(CH₂)₃, Me₂CHCH₂, Me₂CHCH₂, 190-2°/0.5, neg., 0.1; HOCH₂CMe₂, Me(CH₂)₃, Me(CH₂)₃, 155-60°/0.5, neg., 0.1; HOCH₂CMe₂, MeCH₂, Me(CH₂)₃, 170-5°/0.5, neg., 0.1; HO(CH₂)₂₂, Me(CH₂)₄, Me(CH₂)₄, 230-5°/1.0, neg., 0.1; HO(CH₂)₂, C₆H₁₁, C₆H₁₁, - (HCl salt m. 215-16°), neg., 0.1; HO(CH₂)₂, Me(CH₂)₅, Me(CH₂)₅, 194-6°/0.5, 48 min., 0.1. For RN(CH₂CONR1R2)₂ [R, R₁, R₂, m.p. of base or HCl salt (or b.p./mm.), duration in min. and % solution given]: HO(CH₂)₂, Me, PhCH₂CMe₂, 104-4.5°, 25, 0.0005; HOCHMeCH₂, Me, PhCH₂CMe₂, 113-14°, 28, 0.0001; HOCH₂CHEt, Me, PhCH₂CMe₂, 144-5° (HCl salt), 37, 0.0005; HO(CH₂)₃, Me, PhCH₂CMe₂, 164-5° (HCl salt), 82, 0.1; HO(CH₂)₆, Me, PhCH₂CMe₂, 250-60°/0.002, neg., 0.1; (HOCH₂)₃C, Me, PhCH₂CMe₂, 157-8°, 24, 0.001; 2-HOCH₁₀, Me, PhCH₂CMe₂, 108-0.8-5°, 75, 0.0025; PhCHOCH₂, Me, PhCH₂CMe₂, 182-3° (HCl salt), 24, 0.001; PhCHOCHMe, Me, PhCH₂CMe₂, 203-4° (HCl salt), neg., 0.1; HO(CH₂)₂, Me, C₆H₁₁, 190-5°/1.0, neg., 0.1; HO(CH₂)₂, H, PhCH₂CH₂, 72-3° (HCl salt), neg., 0.1; HO(CH₂)₂, Me(CH₂)₃, PhCH₂, 118° (HCl salt), 29, 0.001; HOCHMeCH₂, Me(CH₂)₅, PhCH₂, 195-200°/0.05, 44, 0.1; HOCHMeCH₂, H, 2,6-Me₂C₆H₃, 193-4° (HCl salt), neg., 0.1. For RN(CH₂CONR1R2)CH₂CONR3R4 (R, R₁, R₂, R₃, R₄, b.p./mm., or m.p. of base or HCl salt, duration, and % solution given): HO(CH₂)₂, MeCH₂, MeCH₂, Me(CH₂)₃, 203-5°/1.0, 29, 0.1; HO(CH₂)₂, Me(CH₂)₂, Me(CH₂)₃, Me₂CHCH₂, 198-200°/0.5, 21, 0.1; HO(CH₂)₂, MeCH₂, MeCH₂, Me, PhCH₂CMe₂, 121-2°, neg., 0.1; HO(CH₂)₂, Me(CH₂)₄, Me(CH₂)₄, Me, PhCH₂CMe₂, 92-3°, 63, 0.1; HO(CH₂)₃, Me₂CHCH₂, Me₂CHCH₂, Me, C₆H₁₁, 205-80°/1.0, neg., 0.1; HO(CH₂)₂, Me, PhCH₂CHMe, Me, PhCH₂CMe₂, hygroscopic, 55, 0.001; HO(CH₂)₂, H, PhCH₂CH₂, Me, PhCH₂CMe₂, 158° (HCl salt), neg., 0.1; HO(CH₂)₂, H, PhCH₂CH₂, Me, PhCH₂CMe₂, 42° (HCl salt), neg., 0.1; HO(CH₂)₂, H, Me(CH₂)₅, Me, PhCH₂CMe₂, 260°/1.0, 9, 0.05. For XCH₂CH₂N(CH₂CONR1R2)₂ (X, R₁, R₂, m.p. HCl salt, duration, and % solution): MeCO₂, Me, PhCH₂CMe₂, 169-70°, 32, 0.001; Me(CH₂)₁₀CO₂, Me, PhCH₂CMe₂, 143-5°, 38, 0.01; p-MeC₆H₄CO₂, Me, PhCH₂CMe₂, 168-9°, 42, 0.001; p-O₂NC₆H₄CO₂, Me, PhCH₂CMe₂, 168-9°, 27, 0.0005; MeCO₂, Me(CH₂)₃, Me(CH₂)₃, 212-14°/0.05(base), 32, 0.01; m-C₁C₆H₄CO₂, Me, PhCH₂CMe₂, 87-8° (base, from Me₂COH-petr. ether), active, 0.1; (3-C₅H₄N)CO₂, Me, PhCH₂CMe₂, 158-9°, 35, 0.0005; p-MeOC₆H₄CO₂, Me, PhCH₂CMe₂, 126-7°, active, 0.1; p-H₂N₂C₆H₄CO₂, Me, PhCH₂CMe₂, 199-200°, active, 0.1. Iminoacetamides in which the amido N was derived from aliphatic amines had relatively little local anesthetic action and were more toxic than those derived from aralkyl amines. The use of PhCH₂CMe₂NHMe produced the highest degree of local anesthetic activity in II. Substitution of PhCH₂CHMeNHMe in 1 amide group halved the activity. In the alkanolamine moiety, use of a sterically hindered base (H₂NCH₂CH₂OH) markedly reduced activity. Separation of HO from the tertiary amino group by interposition of CH₂ groups reduced activity. The activity of HOCH₂CH₂CH₂N(CH₂CONMeCMe₂CH₂Ph)₂ was 1/500 of that of the homologous HOCH₂CH₂N(CH₂CONMeCMe₂CH₂Ph)₂. Replacement of HO by NH₂ or Cl, and quaternization of the tertiary amine or reduction of the amide groups to tertiary amines all resulted in nearly complete loss of activity. The activity was not increased by ester formation.

OSC.G 2 THERE ARE 2 CAPLUS RECORDS THAT CITE THIS RECORD (2 CITINGS)

L3 ANSWER 47 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
 AN 1955:19886 CAPLUS
 DN 49:19886
 OREF 49:3870h-i,3871a-i,3872a-i,3873a-b
 TI Organic lithium derivatives
 AU Thiec, Jeanne
 SO Ann. chim. (Paris) (1954), 9, 51-96
 DT Journal
 LA Unavailable
 AB The carbonation of PhLi (I), o-MeC₆H₄Li (II), 2,4,6-(Me₂CH)C₆H₂Li (III), 2,3,5,6(Me₂CH)C₆H₂Li (IV), 2,5-(Me₃C)C₆H₃Li (V), and bornyllithium (VI) (containing about 20% isobornyl-lithium) from pinene-HCl (VII) has been studied by 2 methods. The oxidation of VI with O and the reaction of VI with various ketones were also investigated. Method A for the carbonation of the organo-Li compds. consisted in passing a stream of dry CO₂ over the surface of the Li derivative in dry Et₂O; in method B the solution of the Li derivative was introduced through a nozzle into an atmospheric of CO₂. I prepared from 1/6 mol PhBr in Et₂O (1 mol/1300 cc.) gave by method A at 17-30° 58% PhBz and no BzOH. I (0.1443 mol) in Et₂O (1 mol/4900 cc.) yielded by method B at 0° 72% BzOH and 8% Ph₃COH (VIII); 1 mol I in 1300 cc. Et₂O at 0° yielded similarly 31% BzOH and 33% VIII; and 1 mol I in 4900 cc. Et₂O at 16-19° gave 32% BzOH and 35% VIII. The reaction by method A can be expressed by the equations: 2I + CO₂ → Ph₂C(Oli)₂ (IX); IX + CO₂ → BzPh + Li₂CO₃; whereas by method B in the presence of an excess CO₂ at all times, the reaction seems to proceed according to the equations: I + CO₂ → BzOLi; BzOLi + I + CO₂ → BzPh + Li₂CO₃. II (1 mol) in 1300 cc. Et₂O gave by method A at room temperature 30% o-MeC₆H₄CO₂H (X); in 4900 cc. Et₂O by method B the yield of X was 69%. 2,4,6-(Me₂CH)C₆H₂Br (XI), b₁₈ 146-8°, mixed with 20% PhBr to initiate the reaction, was converted to 50-73% III (in 1 run 90% was obtained). III, prepared from 1/15 mol XI and 1/75 mol PhBr in Et₂O (1 mol/1300 cc. Et₂O) carbonated by method A at 17-25° yielded 53% 2,4,6-(Me₂CH)C₆H₂CO₂H (XII), dark yellow solid, m. 178-80°, and 32% recovered XI. III, prepared from 1/12 mol XI and 1/60 mol PhBr in Et₂O (1 mol/4900 cc.) gave by method B at 0° 87% XII, white solid, m. 185-7°, and 15% recovered XI, and at 14-16° in a similar run 76% XII, m. 176-9°. The steric effect in the carbonation of III leads to the formation of 53% XII by method A, whereas I under the same conditions did not give BzOH. Since no BzOH was formed in the runs with III from the added PhBr, the formation of the III proceeds apparently by the reaction of XI with I. 1,2,4,5-C₆H₂(CHMe₂)₄ (XIII), m. 118°, was prepared in 65% yield from com. iso-PrOH and C₆H₆ in the presence of 20% oleum by the method of Kirrmann and Graves (C.A. 29, 2516.8). XIII treated in CHCl₃ in the cold with Br without a catalyst and the mixture distilled gave at 160-80°/20mm. 40% 2,3,5,6-(Me₂CH)C₆HBr (XIV), needles, m. 148-9°. Li did not react with XIV in Et₂O or petr. ether. XIV (0.03 mol) added to 0.075 mol BuLi in petr. ether (1 mol/2600 cc.), and the mixture refluxed 5 h. and carbonated 24 h. by method A at 17° gave 4.26 g. (52%) 2,3,5,6(Me₂CH)C₆HCO₂H (XV) and 30% mixture of XIII and XIV, m. 100°. XV sublimed at about 100° and melted about 240° (hot Maquenne block). p-C₆H₄(Me₃)₂ (XVI), m. 75-6°, was prepared from Me₂CHCH₂OH and C₆H₆ in the presence of 30% oleum or 80% H₂SO₄. XVI in CCl₄ treated with an equivalent amount of Br in the presence of Fe gave 56% recovered XVI and 27% 2,5-(Me₃C)C₆H₃Br (XVII). XVI treated similarly with 1.75 mol equivs. Br gave a colorless liquid product, b₁₈ 154-6°, consisting of 82% XVII and 18% di-Br derivative which was used for the preparation of V. BuLi (1.5 mol)

in Et₂O treated with 1 mol crude XVII in Et₂O and the mixture refluxed 0.5 h. and analyzed showed 92% organo-Li derivs.; after 2.5 h., 87%; and after 5 h. only 48%. Crude XVII (12 g.) treated with 17.3/269 mol BuLi in 102 cc. Et₂O, and the mixture refluxed after 0.5 h. and distilled gave 3.5 g. (57%) BuBr and 5 g. (59%) XVI, m. 72°. Crude XVII (1 mol) added to 1.5 mol BuLi in Et₂O and the mixture refluxed 0.5 h. gave a solution containing 59% V.

XVII (0.05 mol) treated with 0.075 mol BuLi in 95 cc. Et₂O and the mixture carbonated by method A at about 18° yielded 54% crude 2,5-(Me₃C)₂C₆H₃CO₂H (90% yield based on a 59% yield of V), m. 100-10° (recrystd. from aqueous EtOH and then petr. ether, m. 120-3°; purified through the NH₄ or the NMe₃ salt, m. 128°). Pinene, $[\alpha]_{D}^{25} 7817 - 38.6^{\circ}$, $[\alpha]_{D}^{25} 54617 - 43.5^{\circ}$, was converted to VII, m. 120-3° (from EtOH), $[\alpha]_{D}^{25} 78 - 34.7^{\circ}$, $[\alpha]_{D}^{25} 546 - 38.8^{\circ}$ (c 0.017, EtOH). For the preparation of VI, VII was distilled in a small amount of PhMe up to 110° to remove the EtOH of crystallization VII treated in petr. ether (b. 35-40°) under N in the presence of a crystal of iodine with Li gave 49-54% VI. VII and 10 mol % BuCl added in 1 portion to the Li in petr. ether yielded 75% VI. VI (0.051 mol) (52% yield from VII) in Et₂O (1 mol/4900 cc.) carbonated by method B yielded 51% essentially pure camphane carboxylic acid (XVIII), $[\alpha]_{D}^{25} 78 - 9.6^{\circ}$, $[\alpha]_{D}^{25} 546 - 10.2^{\circ}$ (c 0.025, PhMe). The yield of XVIII in a similar run but by method A (1 mol VI/4 l. petr. ether) was 64%; the crude XVIII yielded 43% pure XVIII, $[\alpha]_{D}^{25} 78 - 11.2^{\circ}$, $[\alpha]_{D}^{25} 546 - 12.8^{\circ}$ (c 0.025, PhMe). VI treated at room temperature with CO₂ under slight pressure (about 50 cc. H₂O) yielded 35.5% XVIII, $[\alpha]_{D}^{25} 78 - 1.8^{\circ}$, $[\alpha]_{D}^{25} 546 - 2.4^{\circ}$, and 17% dibornyl ketone (XIX), b₁₆ 178-80°, n_D¹⁸ 1.502, $[\alpha]_{D}^{25} 78 - 33.7^{\circ}$ (c 4%, EtOH). The partial carbonation of VI during 3 min. with pure CO₂ and with a CO₂-N mixture gave in 4 runs XVIII with values of $[\alpha]_{D}^{25} 78$ between -3.6 and 9.2°. The yield of XIX from VI carbonated only to the extent of 50% was 22%, along with 30% XVIII. VI treated with CH₂:CHCH₂Br and then carbonated gave XVIII with $[\alpha]_{D}^{25} - 10.2^{\circ}$, indicating that no preferential attack of the bromide on the bornyl derivative occurred as with the Grignard derivative of VII. The crude XVIII obtained in these runs, recrystd. from HCO₂H, gave the pure acid as needles, m. 78-80° $[\alpha]_{D}^{25} 78 - 7.5^{\circ}$; recrystd. from PhMe, m. 82-4°, $[\alpha]_{D}^{25} 78 - 4^{\circ}$. The color reaction of Gilman and Schulze (C.A. 19, 2443) for organometallic derivs. with Michler's ketone was also very sensitive with VI, and in general with compds. containing the Li attached to C, whereas compds. with the COLi grouping did not give the color test. VI (0.1067 mol) added dropwise under N to 0.0534 mol HCO₂Et in a little petr. ether, and the mixture refluxed 4 h., hydrolyzed, and distilled yielded 8.1 g. (50%) dibornylcarbinol, b₃₃ 210-20°, which, crystallized twice from C₆H₆-EtOH, gave white crystals, m. 154-60°, $[\alpha]_{D}^{25} 78 - 32.5^{\circ}$, $[\alpha]_{D}^{25} 546 - 34.5^{\circ}$, $[\alpha]_{D}^{25} 436 - 53^{\circ}$ (c 0.02, PhMe). VI (0.108 mol) in petr. ether (1 mol/3380 cc.) agitated under pure O absorbed 80% of the theor. value; the mixture hydrolyzed, the solvent removed, and the residue steam distilled yielded 7 g. product, m. 175-85°, $[\alpha]_{D}^{25} 78 - 30.33^{\circ}$, $[\alpha]_{D}^{25} 546 - 35.66^{\circ}$, $[\alpha]_{D}^{25} 436 - 79.66^{\circ}$, $[\alpha]_{D}^{25} 436 / [\alpha]_{D}^{25} 78$ 2.626, consisting of 35% borneol (XX), 22% isoborneol (XXI), and 43% camphor (XXII). VI (0.124 mol) in petr. ether (1 mol/3400 cc.) agitated at 0° under O without previous removal of the N from the apparatus gave 47% product, m. 192-202°, $[\alpha]_{D}^{25} 78 - 19.0^{\circ}$, $[\alpha]_{D}^{25} 546 - 21.3^{\circ}$, $[\alpha]_{D}^{25} 436 - 41.66^{\circ}$, $[\alpha]_{D}^{25} 436 / [\alpha]_{D}^{25} 78$ 2.19 consisting of 56% XX, 30% XXI, and 6% XXII, which, recrystd. from petr. ether, yielded a mixture, m. 198-203°, of 69% XX, 26% XXI, and 5% XXII. VI (0.1268 mol) stirred

at 0° under a low partial O pressure, the apparatus flushed with O, and the mixture further oxidized at room temperature absorbed 58 and 75% O, resp., and

gave 11.5 g. (60%) mixture, m. 175-87°, $[\alpha]_D^{578}$ -28.66°, $[\alpha]_D^{436}$ -34.0°, $[\alpha]_D^{436}$ -75.0°, $[\alpha]_D^{436}/[\alpha]_D^{578}$ 2.61, consisting of 37% XX, 23% XXI, and 40% XXII. On the basis of these expts. the VI appears to be a mixture of 70% bornyl-lithium and 30% isobornyl-lithium, as compared to the 80:20% ratio determined by the carbonation expts. Me₃CCOCl treated with PhMgBr gave 62% Me₃Cbz, b₁₇ 112-14° nD20 1.5103. 1,3,5-C₆H₃(CHMe₂)₃ (1 mol) in glacial AcOH treated with 1.5 mol HNO₃ gave 95.5% 2,4,6-(Me₂CH)C₆H₂NO₂ (XXXIII), m. 72-3°. XXIII (0.1 mol) hydrogenated in 250 cc. iso-PrOH over 40 g. Raney Ni 2 h. at room temperature gave 75% 2,4,6-(Me₂CH)C₆H₂NH₂ (XXIV), b₁₈ 158-60°, nD20 1.5211; HCl salt, white powder, m. 245-7°, insol. in H₂O. XXIV (75.2 g.) treated with 225 g. MeI and 170 g. Na₂CO₃ in 1125 cc. H₂O heated several hrs. on the water bath, the mixture cooled, and the solid product dissolved in Et₂O treated with KOH, and recrystd. gave 81% crude N,N-di-Me derivative (XXV) of XXIV, yellowish crystals, m. 96-8°, which, recrystd., yielded pure XXV, white needles, m. 97-8.5°, forming in Et₂O with dry HCl in Et₂O the HCl salt, m. 170-2°. XXV treated in CS₂ with AlCl₃ and BzCl on the water bath, the mixture hydrolyzed with ice, extracted with Et₂O, the extract washed with aqueous NaOH, dried with CaCl₂, evaporated, and the residue

crystallized from petr. ether or EtOH yielded 25% 3,2,4,6-H₂N(Me₂CH)C₆H₂Bz (XXVI), m. 128-30°. XXIV treated with excess EtI and Na₂CO₃ gave in the same manner 25% 2,4,6-(Me₂CH)C₆H₂NHET (XXVII), white needles, m. 80-2° (from absolute EtOH), yielding in dry Et₂O with gaseous HCl the HCl salt, m. 262-3°. XXVII with BzCl and AlCl₃ in CS₂ yielded 69% 3,2,4,6-EtNH(Me₂CH)C₆H₂Bz (XXVIII), prisms, m. 124-6° (from petr. ether). VI (0.1 mol) in 322 cc. petr. ether treated dropwise with 0.1 mol Me₂CO in 30 cc. petr. ether, and the mixture hydrolyzed after 0.5 h. and distilled gave 4 g. dimethylbornylcarbinol, b₁₅ 112-16°, nD18 1.4885, d₁₈ 0.9585, MRD 59.06, $[\alpha]_D^{578}$ -13.4°, $[\alpha]_D^{546}$ -16.6° (c 5%, EtOH). VI (0.117 mol) gave similarly with Et₂CO 10.5 g. (40%) diethylbornylcarbinol, b₂₅ 152-6° nD17 1.4885, $[\alpha]_D^{578}$ -13.8°, $[\alpha]_D^{546}$ -16.2° (c 5, EtOH). VI (0.116 mol) treated with 14.3 g. Me₃COCHMe₂, b. 134-6°, and the mixture heated 17.5 h. and distilled gave 17.7 g. (60%) isopropyl-tert-butylbornylcarbinol, colorless liquid, b₁₅ 174-8°, nD16 1.5017, $[\alpha]_D^{546}$ 5.3° (c 5%, EtOH). VI (0.062 mol) in petr. ether heated 30 h. with an equivalent amount of (Me₃C)₂CO, b. 148-50° the mixture distilled, and the product, b₁₅ 190°, cooled and recrystd. from EtOH yielded 59% di-tert-butylbornylcarbinol, m. 70-1°, $[\alpha]_D^{546}$ 2.4° (c 1.66%, EtOH). A slight excess of PhAc in petr. ether treated dropwise with VI in petr. ether, and the mixture heated 6.25 h. and distilled twice gave 20% methylphenyl bornylcarbinol, b₂₀ 189-92°, nD14 1.5450, d₁₄ 1.022, MRD 81.5, $[\alpha]_D^{578}$ 13.4°, $[\alpha]_D^{546}$ 14.6° (c 4%, EtOH). Me₃Cbz and an equivalent amount of VI in petr. ether heated 10 min., the mixture carbonated immediately, hydrolyzed with H₂O, let stand 20 h., and the organic layer distilled gave 38% tert-butylphenylbornylcarbinol (XXIX), colorless liquid with a bluish reflex, b₂₅ 210-14°, nD18 1.5388, $[\alpha]_D^{578}$ 10.8°, $[\alpha]_D^{546}$ 12.6°, $[\alpha]_D^{436}$ 22.4°; the petr. ether layer yielded an addnl. 8% impure XXIX, nD15 1.5270, $[\alpha]_D^{578}$ 22.2°, $[\alpha]_D^{546}$ 26°, $[\alpha]_D^{436}$ 47.4°. VI in petr. ether diluted with PhMe, the petr. ether distilled off, the residual mixture treated with 0.5 mol equivalent of XXVI, refluxed 7 h., hydrolyzed with H₂O, the PhMe layer evaporated, and

the residue treated with aqueous HCl, Et₂O, and acidified H₂O, and purified through the free amine yielded 63% XXVI.HCl, m. 245-7°. VI in PhMe treated with 0.25 mol equivs. of XXVIII in PhMe, and the mixture heated 7 h. and worked up in the usual manner gave 83% recovered XXVIII, m. 80-2°; HCl salt, m. 262-3°. These results show that the reaction of VI with the sterically hindered ketones did not occur at the CO group but rather at another functional group. The reactions of the Li-organic compds. are compared with the known reactions of the corresponding Grignard derivs.

L3 ANSWER 48 OF 48 CAPLUS COPYRIGHT 2010 ACS on STN
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 TI Some derivatives of glucazidone
 AU Maurer, Kurt; Schiedt, Bruno; Schroeter, H.
 SO Berichte der Deutschen Chemischen Gesellschaft [Abteilung] B: Abhandlungen (1937), 70B, 1857-61
 CODEN: BDCBAD; ISSN: 0365-9488
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 GI For diagram(s), see printed CA Issue.
 AB The typical aromatic reagents attack glucazidone (I) (C. A. 29, 7986.5) in the pyridine nucleus, the 3-position apparently being the preferred point of attack in all cases. Only 1 substituent is introduced; if the conditions are made more drastic, the ring system is destroyed. Fuming H₂SO₄ gives a yellow monosulfonic acid (II) which, unlike I, is easily soluble in water. Oxidative degradation with KMnO₄ splits off the SO₃H group with formation of quinoxaline- α -carboxylic acid. While II is completely stable in dry form and gives a series of well crystallized salts, in solution it shows an interesting property. When it is allowed to stand in water in the dark, it remains unchanged, but in the light it soon becomes deep red and after some hrs. deposits a dark red, amorphous, alkali-insol. precipitate. The SO₃H group cannot be replaced by HO, for with alkalies II is rapidly oxidized. Even in the absence of air, the product of alkaline fusion rapidly decomp. When, however, II is nitrated the SO₃H group is smoothly replaced by NO₂, giving nitroglucazidone (III), which is also formed by nitration of I. III cannot be reduced to the amine because the whole ring system is very easily perhydrogenated. Catalytic hydrogenation gives an oily distillable octahydro derivative (IV), which, however, is unstable. I is readily halogenated. In the Br derivative (V) the Br is held quite firmly and cannot be replaced by heating with NH₃ or amines at high temps. and under pressure. V reacts with KOH in MeOH, but not homogeneously, and yields a halogen-free product (VI) of unsharp m. p. which is not identical with 3-hydroxyglucazidone. Chloroglucazidone (VII) is obtained from I with SO₂Cl₂. Nitration of V smoothly gives III. When Br reacts on II, there is apparently first 1,3-disubstitution immediately followed by saponification, giving a halogen- and S-free product with 2 HO groups,
 assumed to be 1,3-dihydroxyglucazidone (VIII). VIII is easily soluble in alkalies and the solution eagerly absorbs O₂, becoming dark. The 2 HO groups can be etherified with CH₂N₂ but not acylated. Attempts to benzoylate or acetylate VIII gave amorphous colored products which dyed wool. The reaction with CH₂N₂ also does not proceed quite normally. Along with the etherification of the HO groups, the quinoid character of the ring system also manifests itself. Quinone, as is well known, adds 2 mols. CH₂N₂ at the ring double bonds; in I, the addition occurs at only 1 double bond, as addition at the 1,11-position is sterically hindered. The product (IX) has the structure VIII, unlike I and the 3-HO derivative, reacts readily with PhNNHH₂, giving a compound (X) with the composition, C₁₈H₁₄O₂N₄, of a normal phenylhydrazone. II (20 g. from 20 g. I slowly added with stirring and cooling to fuming H₂SO₄ (20% SO₃) and then heated 1.5 hrs. on the water bath), prisms with 1 H₂O, m. 275°, previously darkening and decomposing K, Na, Ag salts. III (0.5 g. from 1 g. II slowly added to fuming HNO₃), yellow, m. 215°. VIII (2.5 g. from 5 g. II in cold water treated dropwise with 4 atoms Br), faintly yellowish, m. 206°, difficultly soluble in water with acid reaction, soluble in concentrated acids with red color and repprd. by water, instantly reduces cold Fehling solution and KMnO₄ in acetone, gives with Ac₂O in pyridine brown flocks decomposing above 280° and soluble in concentrated H₂SO₄ with deep blue color;

dilution of the solution produces no flocculation but alkalinization results in a sharp color change. From 0.5 g. VIII with 50 cc. of a solution of CH₂N₂ in ether there is obtained after 15 hrs. 0.35 g. of a yellow compound C₁₄H₁₂O₃N₄, m. 186°, soluble in alkalies, while 0.5 g. VIII suspended in 100 cc. ether and allowed to stand overnight with an ether solution of CH₂N₂ from 5 g. MeN(NO)CONH₂ gives IX, m. 99-100°. X, yellow and red crystals, the yellow form becoming red on heating and m. 202°, soluble in concentrated acids with blue color, changing to violet and finally orange on dilution; alkalies give a yellow color. V (6.5 g. from 7.8 g. I and Br in benzene or CHCl₃), yellow, m. 172°, sublimes at 160° in vacuo, forms with MeI at 100° a methiodide, red, m. 194° (decomposition), which gives a perchlorate, golden yellow, m. 230°, deflagrates vigorously when heated on a spatula. 9-Methyl-10-oxobromoglucazidone (1.3 g. from 2 g. V. MeI in water with alkaline K₃Fe(CN)₆), yellow, m. 178°, becomes discolored in the air.

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